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# Thermochemical recovery from the sustainable economy development point of view—LCA-based reasoning for EU legislation changes

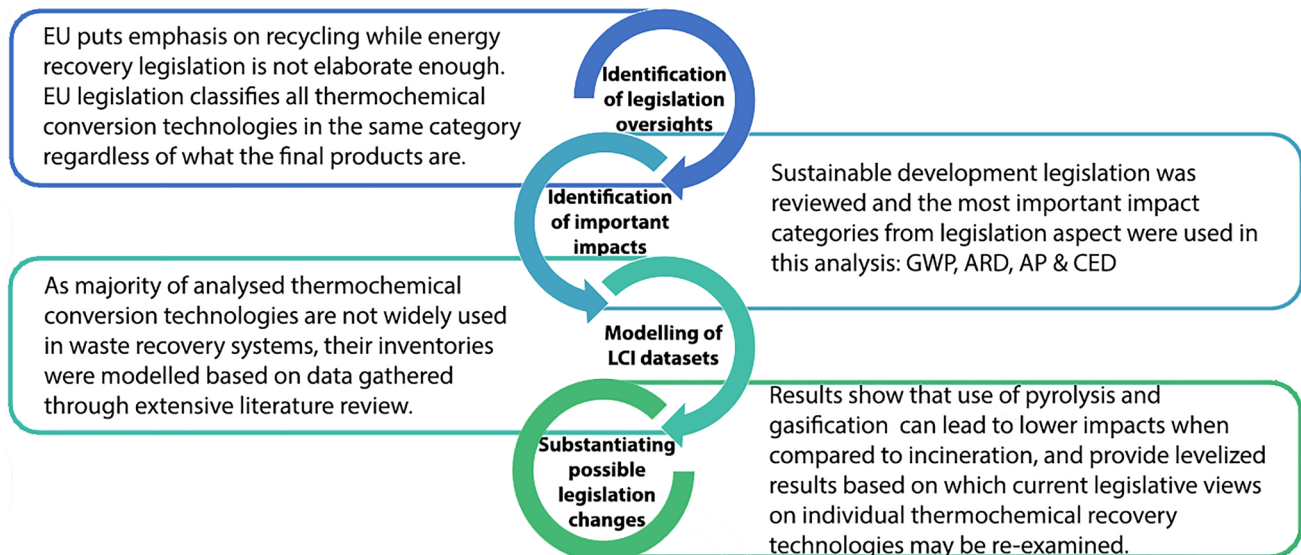
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## Abstract

The EU legislation put the focus on the material recovery of waste while energy recovery is not elaborate enough and all thermochemical conversion technologies are classified in the same category regardless of the final products, which can hamper overall sustainability. Therefore, this research analyses technologies for recovery of plastic waste to review the existing EU legislation and technology classifications. Most important LCA impact categories from the legislation point of view were identified and used in the analysis. As alternative thermochemical recovery technologies are not widely used, their inventories were modelled based on an extensive literature review. Results show that pyrolysis of plastic waste has 46%, 90%, and 55%, while gasification up to 24%, 8%, and 91%, lower global warming, abiotic depletion, and cumulative energy demand-related impacts, respectively, compared to incineration with CHP generation. Incineration-based scenarios show lower impacts only in the acidification potential category which is dependent on energy mixes of substituted energy vectors which are quickly changing due to the energy transition. Thus, alternative thermochemical recovery technologies can help in reaching sustainable development goals by lowering environmental impacts and import dependence. But, before considering new investments, the substitution of less environmentally sustainable fuels in facilities like cement kilns needs to be looked upon. Results of this analysis provide leveled results for environmental and resource sustainability based on which current legislative views on individual thermochemical recovery technologies may be re-examined.

## Graphical abstract



**Keywords** Sustainable development · Legislation changes · Mixed plastic waste · Thermochemical conversion technologies · Environmental and resource sustainability · Life cycle inventory modelling

## Introduction

European production of polymers reached 61.8 million tonnes in 2018, which is equivalent to 17% of the world's production (European Plastics 2019). When the distribution of polymer use by industry sectors is looked upon, 40% of overall production is consumed in packaging production, 20% in the construction sector, 10% in automotive, 6% in electrical and electronic, 4% in household leisure and sports, and 3% in agriculture. Where some products can have a life span of less than a day (such as packaging), others need decades to reach waste streams (like automotive or electronic parts). Therefore, the amounts and composition of plastic waste do not correspond to consumption. Thus, in 2018, from a total of 29.1 million tonnes of collected plastic post-consumer waste, over 61% was packaging waste, although packaging production accounts for 40% of polymers consumption (European Plastics 2019).

Even though polymer waste represents a major problem, until recently there was no dedicated legislative framework on the EU level, and this problem has been only indirectly addressed through non-specific waste legislation. Also, during the years EU put emphasis only on material recovery, while energy recovery of waste is neglected. Because of that, energy recovery technologies have been looked upon mainly from the aspect of mixed waste with the exception of bio-waste. This led to problems with insufficiently elaborated classifications of waste recovery technologies where legislation does not make difference between different thermochemical recovery technologies. This problem is especially pronounced in the case of plastic waste management (WM), especially nowadays the EU put stricter control on plastic waste exports and completely banned exports to non-OECD countries (EP 2020). When all of this is looked at from the plastic WM aspect, where recycling capacity is capped at 30% of production (on a level of 8.5 million tons per year) (Waste Management World 2021), the importance of energy recovery technologies is much more emphasized.

Due to this, this research provides an important contribution by evaluating the environmental impacts of emerging thermochemical technologies for plastic waste valorization, i.e. pyrolysis and gasification, from the points of view of the most actual legislation defined targets, and comparing them with legislatively recognized technologies, with a goal of the revision of the current technology classification and creation of a more sustainable framework. Results of this study could help in reduction in resource use and imports, decoupling prices of petrochemical products and plastic from the oil price, and decrease environmental impacts which leads to

increase in sustainability from an environmental, economic, and political point of view.

## Waste recovery and wider sustainability agenda

The EU principles for MSW management were defined by the Waste Framework Directive (2008/98/EC) through the waste hierarchy and recovery goals which need to be met by 2020. Further along, the New Waste Package (EP 2018) increased targets for MSW reuse and recycling (55% by 2025, 60% by 2030, and 65% by 2035), MSW disposal (max. 10% by 2035), and packaging waste recycling (70% by 2030), as well ban landfilling of separately collected wastes and recyclable/recoverable wastes (from 2030).

One of the waste categories that had a separate legislative framework for many years now is packaging waste—from 1985 and the Directive on containers of liquids for human consumption (85/339/EEC). Over the years, packaging-related guidelines have been adapted to ensure greater environmental protection and set minimum recovery rates, which included incineration, for overall packaging waste, with specific targets by different materials. Based on a review of waste legislation conducted in 2014, EC revised the Directive on Packaging and Packaging Waste (2015/720) and defined measures for the reduction of the consumption of lightweight plastic bags with a thickness below 50 microns. The latest amendment from 2018 under the Waste Package (EP 2018) raised the packaging recycling target to 70% by 2030, with specific targets per material, whereas for plastics it is set to 55% by 2030 (50% by 2025).

Although the packaging and MSW legislations partially covered the plastic WM, only in recent years, it has been actively addressed. European Strategy for Plastics in a Circular Economy (EC 2018a) from 2018 seeks to change how plastic products are designed, manufactured, used, and recycled. Sorting and recycling capacities are to increase fourfold from 2015 to 2030, exports of poorly sorted plastic waste are to be phased out, all plastic packaging needs to be recyclable by 2030, and the use of single-use plastic and microplastics need to be limited. Directive (EU) 2019/904 on the reduction of the impact of certain plastic products on the environment bans disposable plastic products from the market where alternatives are readily available and affordable and limits the use of other plastic products. Targets of 90% separate collection of plastic bottles by 2029 (77% by 2025), 25% share of recycled plastics in PET bottles by 2025, and 30% in all plastic bottles by 2030 were defined.

WM legislation is a constituent part of wider legislation packages that have a goal of solving the problem of energy

and material scarcity in Europe, which at the same time represents economic, political, and security problem of the EU (Tomić and Schneider 2020). Energy scarcity, especially fossil fuels scarcity, and climate change problems are tackled within the same legislation frameworks—the 2020 Climate and Energy Package (EC 2008a) and the 2030 Climate and Energy Framework (EC 2014) whose goals are in line with the Roadmap for moving to a competitive low-carbon economy in 2050 (EC 2011a), the Transport White Paper (EC 2011b), and the Energy Roadmap 2050 (EC 2011c). This path includes GHG emissions reduction of 80% by 2050 (compared to 1990)—transport sector emissions reduction by 60% by 2050 using biofuels and electrification, the power sector should become carbon neutral and heating should be based on renewable electricity or low-emission source. These goals are not specifically connected to EU legislation, as CO<sub>2</sub> emissions mitigation is also part Clean Development Mechanism of the Kyoto Protocol and the United Nations Framework Convention on Climate Change (UNFCCC) (Alizadeh et al. 2014). Along with this path, Heat Roadmap Europe (Persson et al. 2014) classifies waste as the primary district heating heat source. On the other hand, material scarcity is tackled through the Raw Materials Initiative (EC 2008b) and the Flagship Initiative for a Resource Efficient Europe (EC 2011d) which outlines the transformation of the EU economy into a sustainable one till 2050. It emphasizes the importance of decoupling resource consumption (material and energy) and environmental impact from economic growth. Resource Efficient Europe (EEA 2019) strategy aims for a reduction in raw material consumption, an increase in security of supply, support combat against climate change, and limits the environmental impact associated with the exploitation of resources. On this path, the “transformation within a generation—in energy, industry, agriculture, fisheries, and transport systems” is outlined in the Roadmap to a Resource Efficient Europe (EEA 2019) and Circular Economy (EP 2018) is emphasized as the best concept for this transformation. All these plans and aspirations are concise under the Circular Economy strategy and the European Green Deal with initiatives that cover the entire life cycle of products, aiming to ensure that the used resources are kept within the EU economy for as long as possible, and striving to establish climate-neutral Europe.

As it can be seen, EU waste legislation put emphasis on material recovery (i.e. recycling) while energy recovery is subordinate to it and/or clearly neglected. This is not in line with findings presented in previous publications where it is found that implementation of thermolysis-based energy recovery technologies, besides mechanical recycling, is technically and energetically feasible (Mastellone 2019), and that, next to material recovery, energy recovery also represents an important link in the circular economy (Tomić and Schneider 2022). Thus, material and energy recovery

complement each other. Also, EU legislation does not differentiate waste recovery outside of binary classification on material and energy recovery (except anaerobic digestion), and the only well-defined energy recovery technology is waste incineration (Tomić and Schneider 2018). In this context, SUSCHEM (2018) provided an insight into the (thermo)chemical recycling of waste plastics. Post-consumer plastic waste contains impurities and additives (e.g. pigments, paints, and fabric softeners) and other materials (e.g. cellulose, aluminium, and lead), and despite precise selection and separation the polymer materials that enter mechanical recycling are made up of a different mixture of polymers which affects the value and restricts potential use of the recycled material (Ragaert et al. 2020). Also, there is a problem with the quality of the multiple times recycled materials. Other solutions such as thermochemical recycling can be applied to a wide variety of plastic wastes that are not suitable for mechanical recycling and can be the most appropriate recovery technique for mixed plastic waste (MPW). While it can also be sensitive to contaminants of batches with macroscopic contaminants (metal parts, minerals, etc.) and chemicals (chlorine, oxygen, and nitrogen), thus separation of feedstock must be carried out, it is much less sensitive to mixing of different polymers and the majority of contamination-related problems can be solved through the use of catalysts and purification of semi-products/products. Also, mechanical recycling limitations, due to the increase of residues with each new cycle, do not apply to (thermo)chemical recycling (Business Europe 2019). Thus, it represents an option for recycling of mixed and multi-layered, as such, it is complementary to mechanical recycling, and from a life cycle standpoint represents a more viable alternative to incineration and disposal.

Products of alternative thermochemical conversion processes, such as pyrolysis and gasification, can be used as raw materials for fuels, chemicals, and materials production, thereby reducing dependence on petroleum products as well as environmental impact. This helps in decoupling prices of petrochemical products and plastic from the oil price, which is in line with EU legislation. However, in the EC document Best Available Techniques (BAT) for waste incineration (EC 2018b), these technologies are listed under alternative technologies for thermal waste treatment and therefore are classified as waste incineration technologies, even though their products can be used as feedstock material in a wide range of production processes. Considering that in EU categorizes anaerobic digestion as recycling, due to the production of compost-like digestate, the classification of alternative thermochemical conversion technologies into the category of recycling should be considered, or it should be otherwise differentiated from waste incineration. Although the EU is very slow when it comes to legislation changes, EU waste legislation already has integrated mechanisms that

can circumvent the strict regulatory implementations. Like ones in the Waste Framework Directive, which defines that potential deviations from the waste hierarchy, which underlies overall EU waste legislation, can be justified through considerations that include impacts on the level of the whole life cycle. Therefore, the same approach can be used to differentiate particular technologies. Based on these two premises, the hypothesis of this research is formed and states that by using a legislatively recognized approach and analysing technologies through an approach that includes considerations of impacts on the level of the whole life cycle, comprehensive and legislatively meaningful results can be obtained and used for substantiating possible legislation changes.

## Literature review and research objective

Due to importance of “closing the loop”, benefits of WM and recovery were analysed from many angles, from separate collection (Schneider et al. 2021) reuse of wastes (Aydin et al. 2017), chemical recycling (Huang et al. 2022), thermochemical recovery (Ongen 2016; Kremer et al. 2021, 2022; Siwal et al. 2021), to energy recovery via incineration (Tomić et al. 2017; Jadhao et al. 2017; Matak et al. 2021). But, when the sustainability of WM is considered, it needs to be analysed at the level of the overall life cycle and is most often conducted through life cycle assessment (LCA), which is a standardized scientific method for assessing life cycle impacts whose framework was adopted through the ISO 14040 and 14,044:2006 standards. Thus, LCA can be used in line with the propositions of the Waste Framework Directive. In addition, the EC emphasized the importance of LCA and classified it as “the best framework for assessing the potential environmental impacts” (Lima et al. 2018). Therefore, over the past two decades, many LCA of MSW WM systems have been conducted (Istrate et al. 2020), but if the search is limited to recent plastic waste-focused ones, the number of publications is much lower.

Aryan et al. (2019) conducted an LCA of landfilling, recycling, and incineration of PE and PET waste in India using the University of Leiden CML method is conducted. The environmental and economic impacts of recycling, incineration, and landfilling as end-of-life management options for HDPE products were compared using the Eco-Indicator 99 (EI99) LCIA method by Simões et al. (2014). Environmental impact analyses of post-consumer and industrial PLA waste mechanical recycling, chemical recycling as well as thermal treatment were conducted by Maga et al. (2019) and reported results of 11 arbitrary selected midpoint ReCiPe impact categories and the Cumulative Energy Demand (CED) method. Zhang et al. (2020) conducted an LCA and life cycle cost (LCC) analysis of recycling of PET and production of blankets using the Shandong University SDU method and reported results for all 15 midpoint impact

categories. Nakem et al. (2016) used CML and Eco-indicator 99 methods to assess global warming potential (GWP) and energy use in PVC WM. As can be seen, all these researchers focus on only specific, separate, monopolymers recovery, which is the best possible scenario when polymer waste recovery is analysed.

Cascone et al. (2020) analysed plastic granule production from greenhouse covering films through footprint and CED analyses. Ahamed et al. (2020) conducted an LCA of pyrolysis of flexible plastic packaging with pyrolytic oil and nanotubes production and reported on 8 selected ReCiPe midpoint categories. Hou et al. (2018) presented complete BEES method results and compared the environmental impacts of incineration and landfilling as end-of-life treatments for plastic films. Horodytska et al. (2020) used the IMPACT 2002 + method for printed plastic films recycling environmental assessment (upcycling and down-cycling) and compared it to incineration. Lin et al. (2022) analysed the environmental impacts of treatment and recycling of express delivery packaging waste via C-footprint assessment. Beigbeder et al. (2019) analysed end-of-life scenarios (mechanical recycling, incineration, and industrial composting) of polymer (PP and PLA) biocomposites using arbitrary selected 6 midpoint ReCiPe categories. La Rosa et al. (2021) used ReCiPe endpoint and CED results for environmental assessment reporting on chemical recycling of carbon fibre thermosets for the production of thermoplastic composites and compared open and closed-loop scenario results. These researchers analysed the treatment of specific polymer wastes, and obtained results were compared with results for only a minority of available alternative recovery technologies.

Less specific plastics waste streams analyses are even less represented, especially when treatments in different technologies are compared. Thus, Khoo (2019) used the ReCiPe method for reporting climate change, terrestrial acidification, and particulate matter formation results and compared MPW recovery systems consisting of a mix of technologies for energy recovery (thermal treatment with electricity generation, gasification with ethanol production, and pyrolysis with diesel production), but only specific scenarios are analysed without analyses of the influence of alternative products production. Gear et al. (2018) used the CML method for designing MPW thermal cracking process, and compared different system configurations results with incineration and landfilling results, but this is a more specific application of LCA. Cossu et al. (2017) analysed different technologies for the treatment of residual waste from plastic waste separation using the EASYWASTE model. In that case, analysed the waste stream consisted of 57% of plastic (where the rest are metals (27%), textiles (3%), and bio-waste (13%)), while analysed technologies are incineration in different plants (including the substitution of coal in cement kiln),

gasification, and landfilling. While reviewed research analysed substitution of primary fuel in cement kiln as a treatment option, related changes in emissions were neglected. Also, Benavides et al. (2017) analysed fuel production via gasification of non-recycled plastic waste using the GREET model. In this research, the consumption of fossil energy and water is tracked as well as greenhouse gasses production, but only from one technology. Jeswani et al. (2021) compared environmental impacts of households' MPW chemical recycling and energy recovery via pyrolysis using arbitrarily selected midpoint indicators from two different impact assessment methods (Environmental Footprint and ReCiPe). As it can be seen, these publications analyse the treatment/recovery of MPW or (in majority) plastic containing waste streams, but compare them with only arbitrary selected technologies/scenarios or ignore some of the problems connected with modelling of analysed solutions, as well as possible alternative products.

In many cases, simpler and more practical forms of life cycle-based analyses should be used instead of complete, comparative, LCA of systems and technologies (Petrov 2007), which also represent an important mean to overcome prejudice about the complexity of LCA as well as the difficulty in understanding the obtained results by a broader group of people as well as decision-makers. In this context, energy indicators are used in a wide range of activities (Huijbregts et al. 2010; Arvidsson et al. 2012; Scipioni et al. 2013) to identify possible areas for improving production performance or to compare different scenarios during decision-making. Also, Bueno et al. (2015) concluded that “comparisons of alternative systems in terms of direct energy recovery or direct material recovery should be avoided in favour of other indicators already proposed in the LCA framework, such as the CED category from Ecoinvent, or the global warming potential and the Abiotic Resources Depletion categories from the CML 2001 method”. This is based on the properties of those methods, which allow comparison of life cycles of very different systems that encompass energy as well as material flows of a very different nature that are not directly comparable nor can be directly substituted with each other.

CED is an energy-based LCA indicator (Rohrlich et al. 2000) that is quantitative and captures all energy flows which affect the overall life cycle (Huijbregts et al. 2006). It is also an intermediary for environmental impact assessment, correlates with more complex single score impact assessment methods (Mert et al. 2017), gives convergent results with other indicators (such as Ecological Footprint, Cumulative Exergy Extraction in the Natural Environment, Climate Footprint, Ecological Scarcity, and Eco-Indicator), and provides a comparable ranking of impacts (Huijbregts et al. 2010). For this reason, CED is used for selecting a more environmentally friendly alternative (Penny et al.

2013), evaluating the results of overall LCA (Röhrlich et al. 2000), constructing economy-sustainability connection of WM systems (Tomić et al. 2022), and represents an appropriate decision-making tool (Giugliano et al. 2011). Thus, in WM analyses CED was used for sustainability analysis of energy recovery of waste through energy return indicator (Tomić and Schnieder 2017), comparison of municipal WM systems in two towns (Kaufman et al. 2010), and was reported next to CML 2001 results for comparison of different WM practices (Giugliano et al. 2011). Very few publications used CED as an indicator in plastic waste recovery sustainability assessments (Antelava et al. 2019), and only three more recent publications in this field are found—CED results were reported next to Carbon and Water Footprints for energy and environmental assessment of material recovery of greenhouse covering films (Cascone et al. 2020), as well as next to ReCiPe results for the analysis of recycling and incineration of waste PLA (Maga et al. 2019) and for environmental assessment of chemical recycling of carbon fibre thermosets for production of carbon fibre thermoplastic composites (La Rosa et al. 2021). Thus, it can be seen that there is a lack of publications that use CED, as a proven decision-making tool, in MPW management/recovery assessments. This research gap has also been addressed through the presented research.

As it can be seen, while many studies analysed energy recovery of plastic waste from the life cycle perspective, there is a lack of recent studies which are not focused on the specific type of polymers and analyse MPW, especially from an energy recovery perspective. This is even more pronounced from decision-making point of view where a clear lack of comparisons of all applicable technologies can be seen. Also, no previous study has been found to take into account legislative goals in the analysis of the sustainability of the plastic waste recovery, and the majority of reviewed studies report results on all impact category indicators within selected impact assessment method, or on only arbitrary selected ones, without any importance assessments or applicable reasoning. It is important to emphasize these research gaps as EC recognized LCA as a tool that could be used for the elaboration of non-compliance with legislative determinants and thus could be also used as a tool for guiding the changes within the EU legislation. Thus, this research makes a step forward in closing the identified research gaps by conducting LCA-based comparison of alternative thermochemical recovery technologies, taking into account different marketable products that can be produced, and other commonly used technologies for recovery and disposal of MPW through impact indicators which results can be directly connected with specific EU goals in the field of sustainable development. This is done to re-examine the actual industry's views, plastics strategy, and existing stances towards the alternative technologies

for thermochemical recovery of plastic waste, thereby substantiating possible changes in the classification of particular technologies within the WM hierarchy, best available techniques reference document for waste incineration, and broader EU waste legislation. Results of this analysis can provide a levelized assessment of environmental and resource sustainability for dedicated and not-dedicated technologies for MPW recovery in the areas which are emphasized as the most important by EU legislation and previously published research, and can give an answer to the following research question: can alternative thermochemical conversion technologies be better option regarding MPW recovery in the overall sustainable and circular economy oriented development. Based on provided answers, current views on individual thermochemical recovery technologies may be re-examined.

## Methods

This research is comparing the environmental impacts of the two most recognized alternative technologies for thermochemical conversion of mixed polymer waste, i.e. gasification and pyrolysis, with the most commonly used energy recovery and disposal technologies. The results of this research do not include a comparison with material recovery/recycling technologies because this research puts focus on mixed polymer wastes treatment and does not want to question the position of recycling in the waste hierarchy.

## Goal and scope definition

The goal of this research is to use LCA as a legislatively recognized tool to assess the environmental sustainability of differentiation of waste recovery technologies which are by EU legislation classified in the same category, i.e. thermal treatment technologies. Even though the results of this analysis are used to question a part of the EU legislative framework, to reduce the level of aggregation and number of assumptions due to geographical variability, case studies are developed on the basis of the capital city of the newest EU member state (City of Zagreb, Croatia). Croatia became an EU member in 2013, and, since then, implemented many changes in its legislature as well in the WM system to meet EU goals (Luttenberger 2020). Today, the majority of municipal plastic waste is collected as a part of separate packaging waste collection system (Fig. 1). Packaging waste composition is analysed based on 12 samples collected during one day in October of 2019 from different trucks which have collected packaging waste from different parts of the town. Around 120 kg of sampled waste was then homogenized and quartered until the final sample of 7.4 kg was obtained for separation and composition analysis.

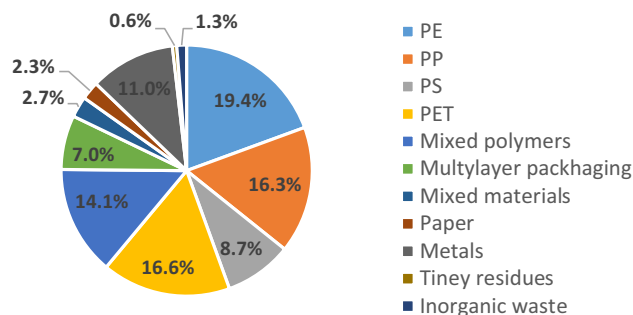
Separation and composition analysis is done by manual separation using Resin Identification Code (RIC) system labels, through examination of material properties (physical properties, melting range, flame tests, and gravity tests).

LCI datasets, that describe analysed WM technologies, are modelled to represent average technology data for corresponding plants for the treatment of one tonne of collected mixed packaging waste of similar properties as one collected in the City of Zagreb, while background processes are modelled through local market activities as described in Ecoinvent database.

LCA is designed per ISO 14044 standard as cradle-to-grave analysis, and ecomaps all activities needed for treatment of generated plastic waste which is separately collected, starting from its generation through collection/transport, pretreatment (i.e. separation, drying, and shredding), and final treatment, which is important to reassess the classification of particular thermochemical recovery technologies from an environmental sustainability standpoint. Due to emphasis on the comparison of technologies for recovery of MPW fraction, analysed systems are made only of essential components to implement analysed technologies so that their influence on results is minimal, and one tonne of collected waste is used as a functional unit. Thus, only separately collected waste recovery is looked upon and connection to local mixed MSW management system is not modelled.

## Analysed systems and boundaries of the systems

Seven different treatment technologies for MPW were analysed and compared—gasification with electricity and ethanol production (a), pyrolysis with emphasis put on oil production (b), incineration with electricity and combined heat and power (CHP) production (c), thermal treatment via co-incineration in the cement kiln (d), and landfilling (e). System boundaries encompass main treatment technologies, collection, and pre-treatment if needed—Fig. 2.



**Fig. 1** Composition of separately collected packaging waste in the City of Zagreb

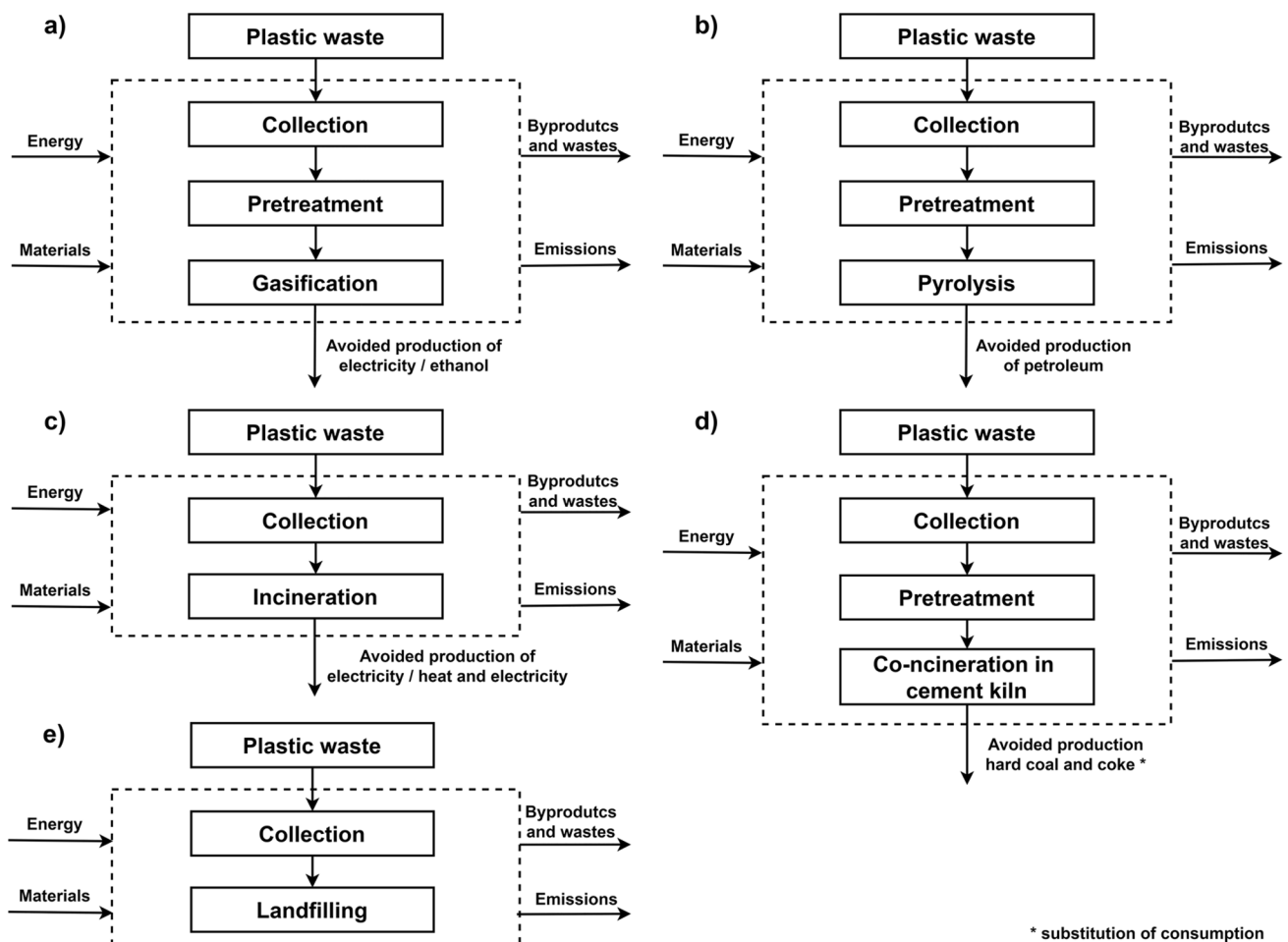


Fig. 2 Boundaries of the analysed systems

Thus, LCA of gasification and pyrolysis encompasses the waste collection, sorting, drying, and shredding of MPW before the main recovery technology. Commonly used technologies such as incineration and disposal usually treat MPW together with other types of wastes (i.e. as it is collected) and pretreatment is not needed, or it is a part of the final treatment plant, as in the case of incineration where separation of metals is done in incineration facility. Regarding co-incineration in cement kiln, because these kinds of plants have strict requirements regarding quality and composition, the collected waste is also sorted, dried, and shredded before use. Gasification can be also used for the treatment of mixed waste, but in this case, this treatment option will not be analysed.

LCA system modelling and uncertainty analysis is done using OpenLCA 1.8.0. software with Ecoinvent 3.5 LCI database where datasets are used for modelling background processes and markets. For final data analysis and presentation of results, Microsoft Excel is used.

### Life cycle inventory (LCI)

Ecoinvent datasets ecomap all known input–output data as data providers allow; thus, it does not incorporate quantitative cut-off criteria (Weidema et al. 2013). To enable consistency, this approach is also applied when using literature data for the creation of inventory datasets; thus, this analysis does not have defined quantitative cut-off criteria. Regarding the possible problems which can arise with using different data sources for technology modelling (Suh et al. 2016), while some of them are avoided by incorporation of all known data in LCI datasets, others are addressed by adaptation to local conditions and matching flows with corresponding local market activities in the Ecoinvent database. Through this, and through averaging of collected datasets, possible problems connected with the use of location-dependent data from different sources, have been also addressed.

Used Ecoinvent database represents one of the biggest commercial LCI databases, and includes average datasets



for all common WM technologies like MPW incineration and waste disposal, but it does not recognize not-so-widely implemented thermochemical conversion technologies like gasification or pyrolysis. To model those technologies, input–output data for plastic waste gasification and pyrolysis technologies are sourced from an extensive literature review, and data for 43 different plants are shown in Tables A1, A2, A3, and A4 in Appendix. To model the average technology life cycle inventory (LCI) (input–output) dataset, all available data for analysed technology are gathered and final datasets are modelled using average values of significant flows for the same type of technologies.

While basic pyrolysis processes produce pyrolytic oil, synthetic gas, and char, some of the plants from the technology review have in-house post-processing in a form of fractional distillation for the production of different fuels—Tables A1 and A2. To circumvent these differences, final LCI datasets modelled pyrolysis without any post-processing, and, to simplify modelling and analysis, produced pyrolytic oil has been marketed as petroleum (oil) due to similar properties and use options. As it can be seen from the gasification technology review results (Tables A3 and A4), it is a most common practice to use produced synthetic gas, which is the main product of the plant, to locally generate electricity. The second most common transformation of synthetic gas is its use for ethanol production which is modelled by (Haig et al. 2013).

Based on literature review data and previous elaborations, average technology LCI datasets for thermic gasification of plastic waste in fluidized bed reactor with electricity generation and catalytic pyrolysis with pyrolytic oil production are modelled (Tables 1 and 2), and the differential dataset for ethanol production, which shows the difference between gasification with electricity production LCI dataset and the ethanol producing one, is presented in Table 3.

As presented LCI datasets are based on datasets that cover input–output flows of tens of actual plants, it was possible to calculate confidence intervals for the inventory data. As specific input–output data cannot be negative, for probabilistic design lognormal distribution is assumed and the geometric standard deviation is calculated as a measure of dispersion analogously to the geometric mean of the corresponding technology data reported in the Appendix.

LCI dataset for pre-treatment is also adapted from the literature (Arena et al. 2003) (Table 4), while the waste collection is modelled based on collection and transport service data (Spielmann et al. 2007) and Ecoinvent data for waste collection with a 21-ton lorry (Table 5).

As in most cases, plastic waste is incinerated in grate incinerators together with MSW or as unrecyclable plastic waste or refuse-derived fuel (RDF). Because of that, incineration technology is modelled as incineration of MPW in an average MSW grate incinerator with an electrostatic

precipitator based on the existing Ecoinvent LCI unit process (UPR) dataset, and the production of heat and electricity has been adapted through a review of data on existing waste incinerators (ISWA 2017; Tomić et al. 2016). Landfilling of plastic waste is modelled as regulated MSW landfill, as plastic waste is landfilled as a part of the MSW stream, and average (representative) technology is modelled based on data from the used LCI database data.

Cement kilns are also used for the final treatment of many types of burnable wastes that meet certain requirements (Rahman et al. 2013). This makes sense because the replacement of primary fuel enables savings of up to 50 €/t (EcoMondis 2018). In available LCI datasets, a cement kiln is defined as a facility whose main fuels are hard coal and petroleum coke, and its substitution with MPW needs to be modelled. To do this, changes in direct emissions due to co-incineration of MPW are modelled on the basis of stoichiometric calculations and laboratory data (Asamany et al. 2017). These data are obtained from the analysis of changes in emissions of  $\text{NO}_x$ ,  $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{SO}_2$ , volatile organic compounds (VOC), particulate matter (PM)  $< 2.5 \mu\text{m}$ ,  $\text{PM} > 2.5 \mu\text{m}$ , and ash production, due to the substitution of coal/coke fuel (1:1 mixture of coal and petroleum coke by mass) with plastic waste materials—plastic containers, films, expanded polystyrene (EPS), Construction and Demolition (C&D) sourced plastics and textiles. It is found that coal/coke substitution with plastic waste, based on the same energy input, can reduce emissions of  $\text{NO}_x$  by up to 79%,  $\text{CO}_2$  by up to 34%,  $\text{SO}_2$  by up to 99%,  $\text{PM} < 2.5 \mu\text{m}$  by up to 14%,  $\text{PM} > 2.5 \mu\text{m}$  by up to 77%, and increase  $\text{H}_2\text{O}$  emissions in air by 194%. Even though VOC emissions are also analysed, because there were no comparative results for the substituted fuel obtained in the same laboratory conditions, these results are not taken into account. Changes in all other emissions and their confidence intervals are also not taken into account. Based on these calculations, the Ecoinvent clinker production dataset is adapted to correspond to 20% of coal/coke fuel mixture substitution by plastic waste mixture, while substitution of emissions is done by supplied energy equivalent. The derived LCI dataset is shown in Table 6.

The inputs and outputs of the respective technologies are connected with the outputs of other activities from the used database and in a majority of cases market activities (i.e. with LCI datasets for local market activities for particular materials, energy vectors, and/or services). Market activities datasets represent a market mix of all activities with the same reference product in a particular area and include the impacts of all the activities that precede the use of an individual product in a specific location (including production, transportation, processing, and transformation), thus representing the average market data for the particular geographic area.

**Table 1** LCI dataset for gasification with electricity production

		Flow	Unit	Value	$\sigma_g$
Input	Input*	Waste plastic, mixture	t	1.000	1.000
		Energy consumption	Electricity, medium voltage	kWh	524.287
	Other inputs	Oxygen	kg	1170.461	1.128
		Zeolite, powder	kg	53.500	1.000
		Diesel	l	0.209	1.000
		Sodium hydroxide, without water, in 50% solution state	kg	5.000	1.000
		Activated carbon, granular	kg	0.500	1.011
		Feldspar	l	0.417	1.000
		Heat	kWh	146.377	2.089
		Water, turbine use, unspecified natural origin	l	5591.360	1.969
	Lime, hydrated, loose weight	kg	6.469	1.008	
Output	Additional fuel:	Natural gas, high pressure	kWh	1560.000	1.000
		Energy products	Electricity, medium voltage	kWh	1267.587
	Material by-products	Steam	kg	2210.871	1.876
		Refinery gas	kg	214.000	1.000
	Other:	Sulphur	kg	1.500	1.000
		Salt tailing	kg	5.500	1.000
		Ground granulated blast furnace slag	kg	112.000	1.000
		Char, for disposal	kg	148.660	1.000
		Blast furnace slag	kg	7.942	3.653
		Coal tar	kg	141.500	1.000
		Process-specific burdens, residual material landfill	kg	44.462	2.665
		Waste zeolite	kg	1.695	1.000
		Fly ash and scrubber sludge	kg	92.822	2.131
		Refinery sludge	kg	22.500	1.008
		Process-specific burden, sanitary landfill	kg	6.500	1.000
		Output	Emissions in air:	Particulates, >2.5 um. and <10um	kg
Particulates, <2.5 um	kg			3.841E-02	2.425
Carbon dioxide	kg			1899.1783	2.631
Methane	kg			0.4725	3.220
Hydrogen chloride	kg			2.947E-02	2.184
Sulphur dioxide	kg			1.142E-01	1.657
Sulphur oxides	kg			1.010E-01	1.028
Dinitrogen monoxide	kg			9.900E-02	4.052
Nitrogen oxides	kg			7.154E-02	1.146
Carbon monoxide	kg			3.975E-01	3.371
Mercury	kg			9.696E-07	1.738
Cadmium	kg			4.807E-06	3.557
Lead	kg			1.607E-03	4.559
VOC, volatile organic compounds	kg			2.350E-01	4.457
Hazardous Air Pollutants (HAPs), unspecified	kg			5.000E-02	1.000
Ammonia	kg			3.350E-05	1.039
Dioxins and furans, unspecified	kg	5.981E-12	1.299		
Acetaldehyde	kg	0.030	1.000		

**Table 1** (continued)

	Flow	Unit	Value	$\sigma_g$
	NMVOC, Non-methane volatile organic compounds	kg	0.100	1.000
	Antimony	kg	6.562E-04	4.023
	Arsenic	kg	9.594E-07	1.390
	Titanium	kg	2.591E-06	1.270
	Chromium	kg	5.412E-04	2.608
	Iron	kg	2.514E-03	1.876
	Copper	kg	3.322E-03	2.985
	Zinc	kg	6.250E-05	1.000
Emissions in water:	Wastewater	kg	6077.150	2.578

**Table 2** LCI dataset for pyrolysis

		Flow	Unit	Value	$\sigma_g$	
Input	Input*	Waste plastic, mixture	t	1.000	1.000	
	Energy consumption:	Electricity, medium voltage	kWh	283.215	3.554	
	Other:	Zeolite, powder	kg	21.346	2.258	
		Water, turbine use, unspecified natural origin	l	1587.770	3.847	
	Additional fuel:	Natural gas, high pressure	MWh	0.431	2.050	
Output	Energy products:	Synthetic gas	MWh	0.065	1.000	
		Pyrolytic oil	kg	708.653	1.140	
		Pyrolytic gas	kg	142.608	1.523	
	Other:	Char, for disposal	kg	77.805	1.351	
		Process-specific burdens, residual material landfill	kg	128.117	1.602	
		Waste zeolite	kg	15.050	2.175	
		Process-specific burden, sanitary landfill	kg	15.627	3.544	
		Hazardous waste, for incineration	kg	23.000	2.470	
		Wastewater, average	l	613.754	4.797	
		Emissions in air:	Particulates, > 2.5 um, and < 10um	kg	0.078	3.742
			Carbon dioxide	kg	401.445	1.328
			Hydrogen chloride	kg	1.500E-04	1.000
			Hydrocarbons, unspecified	kg	2.058	1.452
			Sulphur dioxide	kg	0.045	4.129
			Dinitrogen monoxide	kg	0.459	1.563
			Nitrogen oxides	kg	0.583	3.144
			Carbon monoxide	kg	0.482	2.013
			Mercury	kg	1.764E-11	1.000
			Lead	kg	5.050E-03	2.595
		VOC, volatile organic compounds	kg	0.273	4.747	
Ammonia	kg	5.500E-03	1.138			

Life cycle impact assessment (LCIA).

However, this research wants to assess the compatibility of analysed technologies with EU legislation goals and challenge

the current classification of energy recovery technologies.

Because of it, the choice of LCIA indicators is steered by findings of an overview of actual legislation frameworks

**Table 3** Gasification with ethanol production—Differential LCI dataset

		Flow	Unit	Value	$\sigma_g$
Input	Other inputs	Water, turbine use, unspecified natural origin	kg	+5322.000	1.969
	Energy consumption	Heat	kWh	+800.000	2.089
Output	Production	Ethanol	kg	584.000	1.667
		Reactor off-gas	kWh	1900.000	1.000
		Electricity medium voltage	kWh	-1454.760	1.620
	Other	Wastewater, average	kg	+5195.000	2.578

**Table 4** LCI dataset for waste pre-treatment

		Flow	Unit	Value	$\sigma_g$
Input	Input*	Waste plastic mixture, unsorted, from collection service	t	1.730	1.000
	Energy consumption	Diesel	kg	1.4E-3	1.105
Electricity, medium voltage		kWh	0.284	3.554	
Output	Output	Plastic waste mixture, sorted	kg	1.29	1.000
	Residues	Municipal solid waste	kg	0.435	1.000

**Table 5** LCI dataset for collection

		Flow	Unit	Value	$\sigma_g$
Input	Energy consumption	Diesel	kg	0.336	1.105
		Other inputs	Road	m·a	0.00064
		Waste collection lorry, 21 metric ton	items	4.520E-7	1.000
Output	Product*	Municipal waste collection service by 21 metric ton lorry	t·km	1	1.000
Emissions in air		Ammonia	kg	7.95E-6	1.221
		Benzene	kg	6.77E-5	1.221
		Cadmium	kg	4.480E-09	2.253
		Carbon dioxide, fossil	kg	1.060	1.000
		Carbon monoxide, fossil	kg	2.730E-3	2.239
		Chromium	kg	1.690E-08	2.253
		Copper	kg	5.710E-7	2.253
		Dinitrogen monoxide	kg	5.250E-5	1.221
		Lead	kg	4.870E-09	2.253
		Methane, fossil	kg	8.460E-5	1.221
		Nickel	kg	2.350E-08	2.253
		Nitrogen oxides	kg	7.58E-3	1.221
		NMVOC, non-methane volatile organic compounds	kg	3.450E-3	2.253
		Particulates, <2.5 um	kg	6.150E-4	1.221
		Particulates, >10 um	kg	1.750E-4	1.221
		Particulates, >2.5 um, and <10um	kg	1.050E-4	1.414
		Selenium	kg	3.360E-09	2.253
		Sulphur dioxide	kg	2.020E-4	1.000
		Toluene	kg	2.710E-5	1.221
Xylene	kg	2.710E-5	1.221		
Zinc	kg	3.330E-6	2.253		

**Table 6** LCI dataset for clinker production with co-incineration of MPW

		Flow	Unit	Value	$\sigma_g$	
Input	Input*	Waste plastic, mixture	kg	0.00597015	1.000	
	Energy consumption	Hard coal	kg	53.500	1.105	
		Heavy fuel oil	kg	0.209	1.105	
		Light fuel oil	kg	5.000	1.105	
		Petroleum coke	kg	0.417	1.105	
	Other inputs	Ammonia, liquid	kg	0.000908	1.105	
		Bauxite	kg	0.00012	1.105	
		Calcareous marl	kg	0.466	1.105	
		Clay	kg	0.331	1.105	
		Industrial machine, heavy, unspecified	kg	0.0000376	1.105	
		Lime	kg	0.841	1.105	
		Lime, hydrated, loose weight	kg	0.00392	1.105	
		Lubricating oil	kg	0.0000471	1.105	
		Meat and bone meal	kg	0.00961	1.105	
		Refractory, basic, packed	kg	0.00019	1.105	
		Refractory, fireclay, packed	kg	0.0000821	1.105	
		Refractory, high aluminium oxide, packed	kg	0.000137	1.105	
		Sand	kg	0.00926	1.105	
		Steel, chromium steel 18/8, hot rolled	kg	0.0000586	1.105	
		Tap water	kg	0.34	1.105	
		Water, unspecified natural origin	m <sup>3</sup>	0.00162	1.105	
		Additional fuel:	Diesel	MJ	524.287	1.105
			Electricity, medium voltage	kWh	1170.461	1.105
			Natural gas, high pressure	m <sup>3</sup>	0.500	1.105
Output	Products:	Clinker	kg	1.00	1.000	
	Other outputs:	Inert waste, for final disposal	kg	0.00008	1.105	
		Municipal solid waste	kg	0.000045	1.105	
Output	Emissions in air:	Ammonia	kg	0.0000228	1.105	
		Antimony	kg	0.000000002	1.105	
		Arsenic	kg	0.000000012	1.251	
		Beryllium	kg	0.000000003	1.251	
		Cadmium	kg	0.000000007	1.251	
		Carbon dioxide, fossil	kg	0.829509391	1.105	
		Carbon dioxide, non-fossil	kg	0.014929192	1.105	
		Carbon monoxide, fossil	kg	0.000472	1.105	
		Chromium	kg	1.45E-09	1.251	
		Chromium VI	kg	5.5E-10	1.251	
		Cobalt	kg	0.000000004	1.251	
		Copper	kg	0.000000014	1.251	
		Dioxins, measured as 2,3,7,8-tetrachlorodibenzo-p-dioxin	kg	9.6E-13	1.105	
		Hydrogen chloride	kg	0.00000631	1.251	
		Lead	kg	0.000000085	1.253	
		Mercury	kg	0.000000033	1.251	
		Methane, fossil	kg	0.00000888	1.105	
		Nickel	kg	0.000000005	1.251	
		Nitrogen oxides	kg	0.001003442	1.105	
		NM VOC, non-methane volatile organic compounds	kg	0.0000564	1.105	

**Table 6** (continued)

	Flow	Unit	Value	$\sigma_g$
	Particulates, < 2.5 $\mu\text{m}$	kg	2.44245E-05	1.105
	Particulates, > 10 $\mu\text{m}$	kg	6.07498E-06	1.251
	Particulates, > 2.5 $\mu\text{m}$ , and < 10 $\mu\text{m}$	kg	8.50067E-06	1.434
	Selenium	kg	0.000000002	1.253
	Sulphur dioxide	kg	0.000328563	1.105
	Thallium	kg	0.000000013	1.251
	Tin	kg	0.000000009	1.253
	Vanadium	kg	0.000000005	1.251
	Water	$\text{m}^3$	0.000300629	1.105
	Zinc	kg	0.000000006	1.251
Emissions in water:	Wastewater	$\text{m}^3$	0.001666	1.221

regarding WM but also regarding the sustainable development of the entire European economy, as well as findings gathered through literature review in the field of WM and recovery (analyses, comparisons, and decision-making), which are provided as a part of the Introduction section. EC emphasized the importance of assessments on the level of the whole life cycle, especially LCA. Because of this, in this research, the CML baseline 2001 problem-oriented impact assessment characterization method is used for conducting overall LCA, which belongs to a group of problem-oriented approaches (mid-point categories) that are used for environmental and human impact assessments (Aryan et al. 2019).

As can be seen from the legislative review, one of the main EU problems is resource scarcity (material and energy), which also encompasses waste recovery, and impact on climate change. Due to this, this research takes into account three CML mid-point category indicators—global warming potential (GWP (expressed in  $\text{kg CO}_{2\text{eq}}$ )), abiotic resource depletion (ARD (in  $\text{kg S}_{\text{beq}}$ )), and acidification potential (AP (in  $\text{kg SO}_{2\text{eq}}$ )). The first two indicators are chosen as they cover emissions of greenhouse gasses and depletion of a wide range of earth resources which is directly connected to EU legislation frameworks. While the World Health Organisation (WHO) emphasizes the positive impacts of the circular economy on GHG emissions, it also comments on the positive influence on air pollution (WHO 2018). Also, in previous publications, the importance of reduction of air pollution in the context of not only EU legislation aiming at improving environmental sustainability and at carbon neutrality, but also international agreements such as the Sustainable Development Goals, Kyoto Protocol, and Paris Climate Agreement is clearly identified (Torkayesh et al. 2021).

Thus, the last tracked indicator covers the emission of air pollutants.

GWP accounts for GHG emissions with a time horizon of 100 years, to account for different release times. It tracks emissions of  $\text{CO}_2$  from fossil sources only and does not account for biogenic emissions. ARD assesses the extraction of metals, minerals, and fossil fuels considering their depletion rate and reserves. AP covers emissions of compounds with acidification potential— $\text{NO}_x$ ,  $\text{SO}_x$ , and ammonia which are considered the main air pollutants by the National Emissions Ceilings (NEC) Directive (2016/2284/EU).

Previous research identified that comparisons of alternative systems in terms of direct energy or material recovery should be avoided in favour of indicators such as CED from Ecoinvent or GWP and ARD from the CML 2001 method (Bueno et al. 2015). Also, CED has been identified as a suitable sustainability indicator for decision-making in WM systems (Röhrlich et al. 2000). Because of that, next to CML 2001 category indicators, this analysis also tracks energy flows (consumption and production) and reports on associated impacts through CED results.

To assess the combined influence of all input uncertainties and a degree of possible deviations of results, especially for modelled pyrolysis and gasification technology results, uncertainty propagations and quantifications, using reported confidence intervals, are reported. For this Monte Carlo approach is used, as the most popular approach for obtaining uncertainty analysis results as a part of LCA (Lloyd and Ries 2007). Normalization and weighting are per ISO standards defined as optional elements of LCA and were not performed as a part of this analysis due to the uncertainties which are associated with the normalization factors calculations (Heijungs et al. 2007; Hung and Ma 2009) as well

as because the associated loss of transparency (Reap et al. 2008).

### Results and discussion

Based on described methods, environmental impact results are calculated using OpenLCA 1.8.0. program—Figs. 3, 4 and 5. The allocation of impacts and benefits of production of secondary material and energy flows (multifunctionality consideration) was performed using the system expansion method and production was valued through the avoided consumption of primary products/resources. In interpreting the results, a negative value indicates the positive effect, and a higher positive value represents the greater adverse impact.

The worst GWP results can be seen for incineration-based scenarios and pyrolysis shows the best results, a similar situation is in the case of ARD with a difference of gasification with electricity production which here show worse results than incineration, and on the other hand, incineration with electricity production shows the best results regarding AP while all other dedicated waste treatment technologies lag at least 20% behind it, and pyrolysis shows the lowest

positive impact regarding AP. Co-combustion of MPW in cement kiln shows overall the best results, being second only to pyrolysis regarding ARD. The last scenario used for comparison, landfilling, shows a relatively small negative impact across all impact analyses which is due to landfilling of inert material and the majority of the impacts come from energy and material consumption which are not offset by any production.

To validate results and compare uncertainties within newly modelled LCI datasets the Monte Carlo Analysis is performed which is a sampling-based uncertainty quantifying method, where, to estimate the uncertainty (i.e. probability distribution of the specific result) the calculation needs to be repeated a number of times (Helton et al. 2006). An obtained probability distribution can be then used for informing decision-makers on characteristics/probability of obtaining reported results through statistical data. There is no clear argument on a number of Monte Carlo runs needed for effective uncertainty analysis, and literature data suggest from 100 iterations (BIPM 2008) over 2000 (Hongxiang and Wei 2013) to over 10,000 (Xin 2006). Thus, in this analysis, Monte Carlo analysis of 10,000 runs is done and statistical analysis is performed on obtained distributions.

Fig. 3 GWP results in kg CO<sub>2</sub>eq

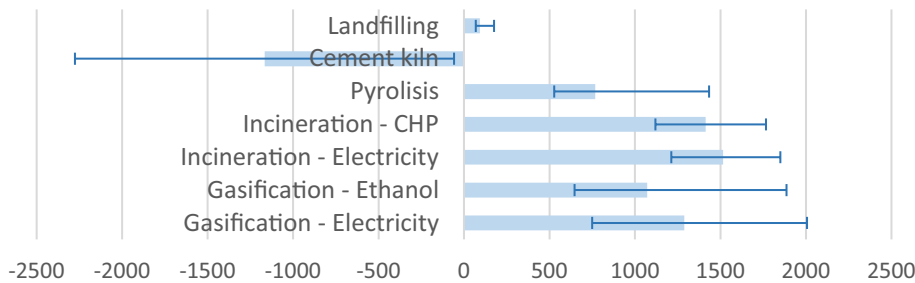


Fig. 4 AP results in kg SO<sub>2</sub>eq

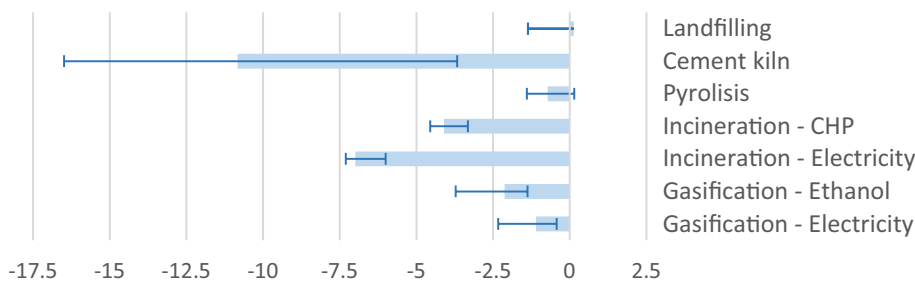
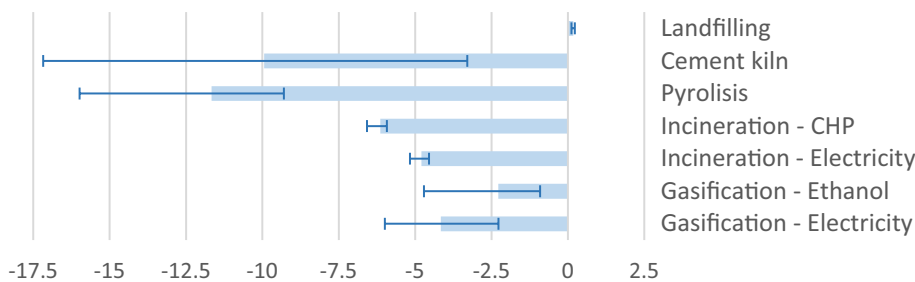


Fig. 5 ARD results in kg Sb<sub>eq</sub>



Following obtained statistical analysis results, 5% Percentile and 95% Percentile results are denoted by corresponding error lines (Figs. 3, 4 and 5) to depict the quality of assessment and compare uncertainties. It can be seen that the smallest deviations are obtained for landfill and incineration-based technologies, which can be expected as these LCI datasets are based on Ecoinvent data. Possible errors in results for pyrolysis and gasification-based scenarios are double on average when compared to incineration-based scenarios, and the biggest possible errors can be expected with waste treatment in cement kiln due to the biggest dataset needed to model this technology. Overall, even though some scenarios show much bigger dissipation of results, there is a small chance that it can affect previously drawn conclusions and rankings.

To analyse the main drivers of these results, the contribution of dedicated technologies and markets are shown in Figs. 6, 7 and 8. To make diagrams more readable, only the six most significant impacts are shown. Here, the greatest overall greenhouse gasses (GHG) emissions are associated with the incineration of MPW with electricity production, followed by incineration with CHP production. This is expected due to direct GHG emissions, which represent the biggest impact, and are only partially offset by energy production. Indirect emissions impacts are at least two orders of magnitude smaller. Gasification-based technologies show better results than incineration-based ones mainly due almost 40% smaller direct emissions. Other significant emissions come from catalyst use and heat consumption. These emissions are partially offset through electricity, steam, and ethanol productions. Pyrolysis has the best results among

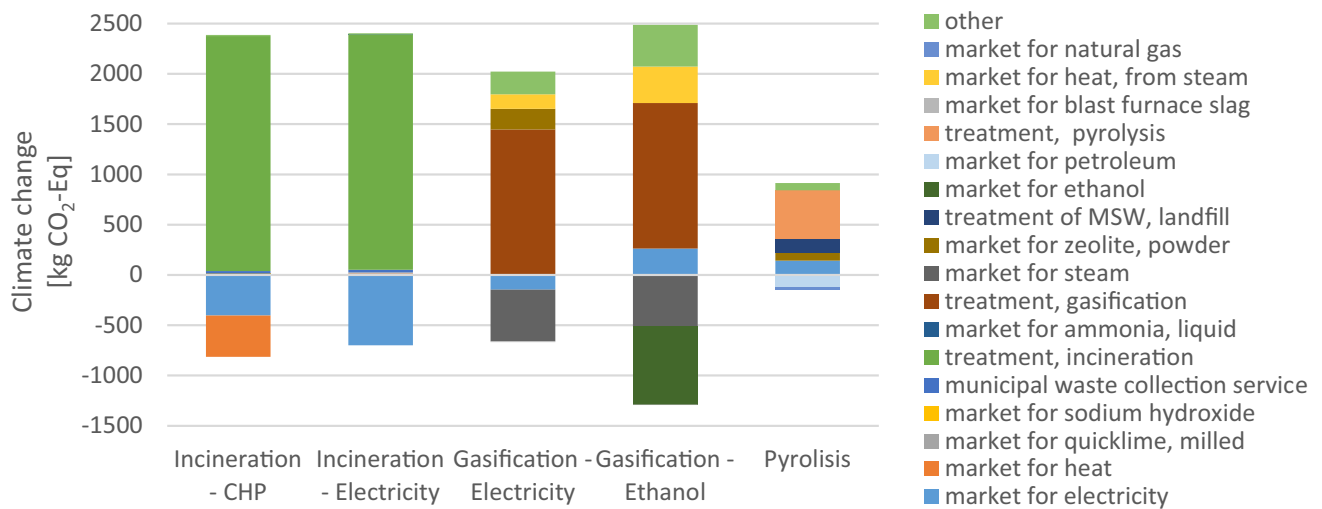


Fig. 6 Climate change—the main contributors

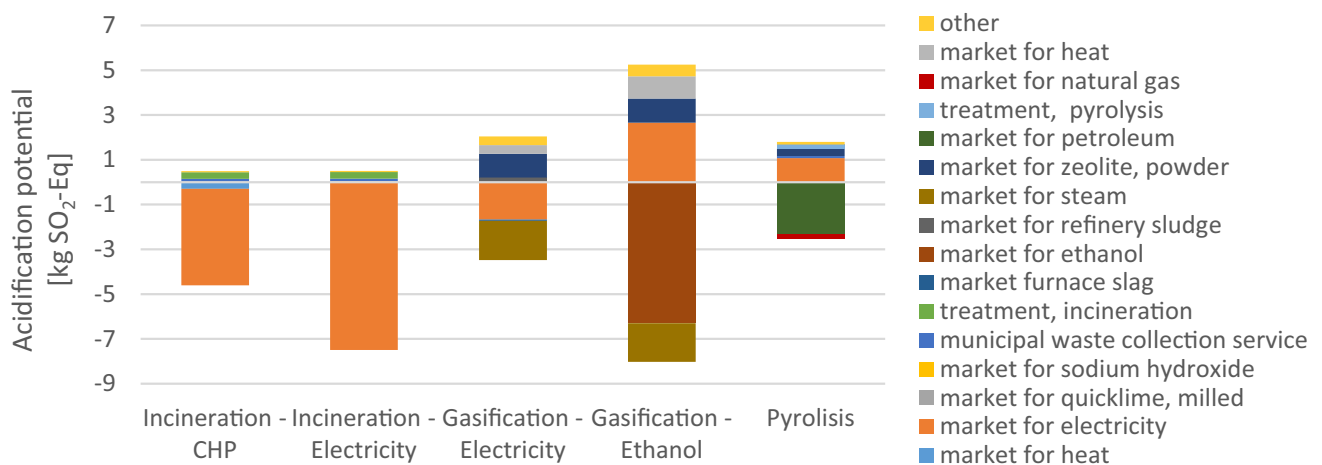
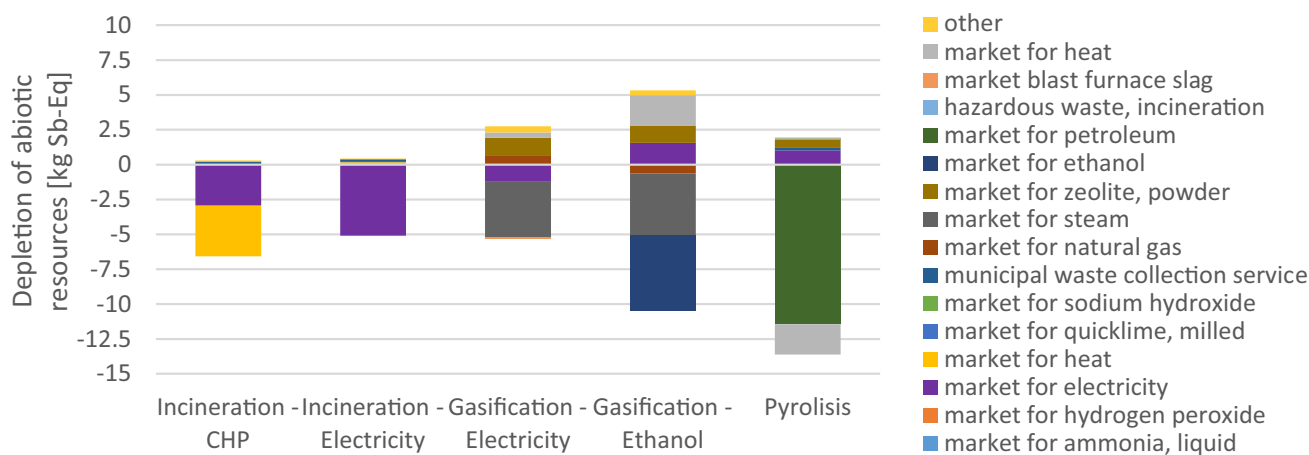


Fig. 7 Acidification potential—the main contributors





**Fig. 8** Abiotic depletion—the main contributors

all recovery technologies due to the smallest direct emissions which are then partially offset with production, mainly pyrolysis oil (which can replace petroleum in refineries). On the other hand, in the case of co-combustion in cement kiln which results are not presented in diagrams because values of influences by each contributor (technology/market) are not in the same order of magnitude as in other scenarios, the majority of GHG emissions are direct emissions, and the majority of emission savings comes from coal and coke substitution. Other impacts are just a few percent and come from the consumption of other inputs needed for clinker production.

Regarding AP, the smallest positive impact of dedicated recovery technologies is recorded for pyrolysis, as negative impacts associated mainly with electricity consumption and catalyst use are marginally smaller than petroleum substitution-connected impacts. For gasification with electricity production, the biggest negative AP impact is from catalyst consumption, followed by energy consumption and disposal of waste products. Gasification direct emissions contribute only to 10% of emissions compared to catalyst consumption. Regarding positive influence, the situation is similar to the case of GWP where ethanol production has a bigger influence than electricity production. Incineration with electricity production shows the best results due to the local electricity mix which has a bigger AP than heat from district heating. On the other hand, due to modern flue gas filtration, direct emissions of waste incinerators are only 2.4 times bigger than those of waste collection services. In the treatment of MPW in cement kiln, there are similar results on the positive side, where clinker produced with alternative fuel in mix offset all acidification-related emissions, but on the negative side, acidification contribution is more dispersed. Thus, around 60% of emissions are direct emissions, while the rest are distributed evenly across heavy fuel oil, electricity, hard coal, and lime consumptions.

Pyrolysis shows the best ARD results that are directly connected to the production of pyrolysis oil which is valued as petroleum substitution and more than makes up for abiotic depletion due to electricity and catalyst consumption. In the case of gasification with ethanol production, ethanol and steam market substitution are two main positive contributors, while negative contributors are catalyst use, electricity, and heat consumption. In the case of electricity production, results are worse due to four times lower positive influence than ethanol substitution on market, regardless of smaller energy requirements on the input side. Regarding incineration, the only significant overall impact on ARD result is due to energy substitution on respective markets, while all other impacts are at least one order of magnitude smaller. The cement kiln shows similar results as before on the impact reducing side, while the main contributors to resource consumption are fuel and energy consumption (coal, fuel oil, and electricity).

As can be seen, AP shows different results compared to the other two impact categories. This is mainly due substitution of electricity with the average local energy mix which leads to bigger acidification impact reduction but also increases burdens associated with non-electricity producing technologies. Also, a relatively big acidification impact is associated with catalyst consumption. Direct impacts have a minor impact here, which cannot be said for the GWP category where direct emissions generally have the biggest impact. On the other hand, the ARD impact category only accounts for material and energy consumption. ARD factor is based on the state of resources, their reserves, and exploitation rate, and is expressed in the form of equivalent of reference resource depletion—antimony depletion. In this form, this characterization factor accounts for material depletion and does not include consumption of resources which overall reserves cannot be estimated, thus neither is renewable energy accounted for.

Overall results show that incineration, when compared to technologies that produce semi-products (ethanol or petroleum), shows substantially worse overall results when all impact categories are looked upon. Deviation of this conclusion can be seen in the case of AP where incineration with electricity production shows the best results. Climate change results are the most influenced by direct emissions, because cracking of hydrocarbons leads to GHG emissions, and avoided emissions cannot compensate because there are more efficient ways for the production of these products. The worse situation is with incineration because complete combustion leads to the biggest emissions on the one side and avoided emissions from electricity or heat production are low because these energy vectors can be produced from many energy sources including renewable ones. Pyrolysis shows one of the best results, mainly because it has the smallest direct emissions due to the production of the heavier main product. At the same time, the only technology with a negative climate change impact is the cement kiln, mainly due to the type of fuel it substitutes, and reduced CO<sub>2</sub> emissions with its substitution. AP results show opposite results regarding incineration mainly due to efficient flue gas filtration/scrubbing, while avoided impacts are energy mix dependent. Other thermochemical transformation technologies have significant negative impacts due to catalyst use and electricity consumption which pushes even the technology with the largest avoided impacts (gasification with ethanol

production) to a third place. Similar results regarding negative impacts can be also seen in the case of ARD but final results differ due to avoided production associated impacts, where the biggest ones are due to ethanol and pyrolysis oil/petroleum production. The market placement of other gasification and pyrolysis products also leads to substantial positive environmental impacts.

Another used LCA-based approach is CED assessment which accounts for the overall consumption of each analysed chain and displays its contributions in a form of consumed primary energy (PE) equivalent—Fig. 9. Thus, the CED result accounts for the consumption of all materials from nature through the energy used for their extraction. Not only that it looks upon energy use through extraction, but also through reprocessing, transformation, production, recovery, and disposal, thus covering the entire life cycle of products and materials, taking into account renewable, fossil, and nuclear energy consumption. Even though it does not account for direct contributions it is used for the overall environmental sustainability assessment of WM and recovery systems.

Regarding PE, gasification with ethanol production gives the best results, followed by pyrolysis while incineration is lagging. As can be seen, even though the CED approach looks into energy and material consumption, its results differ from ARD results. Why that is can be seen in Fig. 10 which shows the contribution per type of energy source.

Fig. 9 CED results in MJ

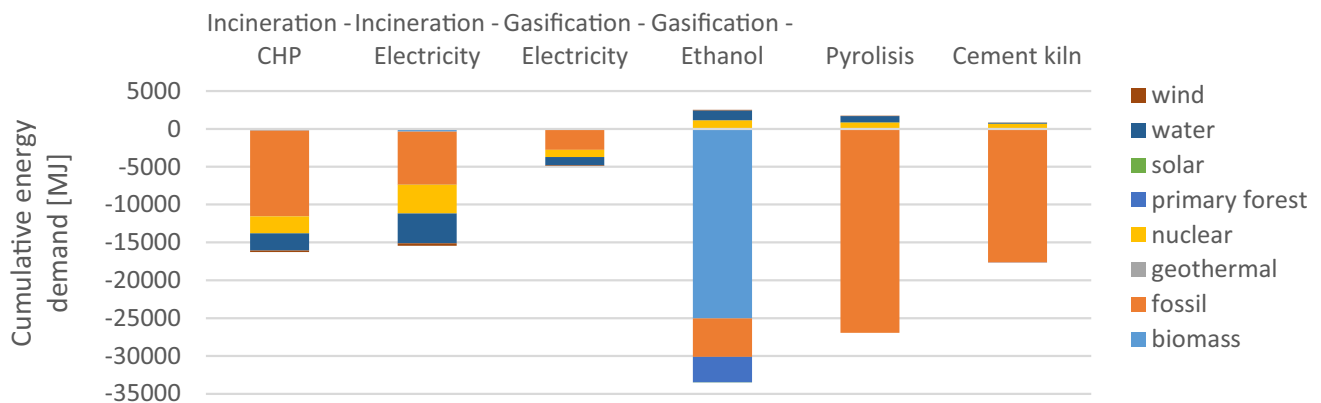
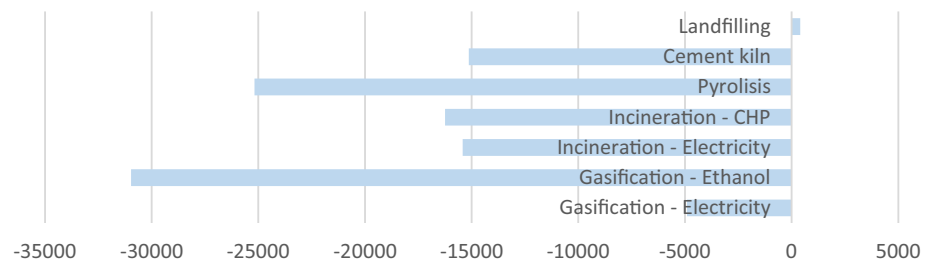


Fig. 10 Cumulative energy demand results per energy source

As it can be seen, 16% of overall PE consumption is covered by renewable energy sources (RES) in the case of incineration with CHP production, 30% in the case of incineration with electricity production, 9% in the case of gasification with electricity production, 3% in the case of pyrolysis production, and 55% in the case of gasification with ethanol production. As ARD, per its definition, take into account resources reserves and exploitation rate, it neglects renewable resources, and thus, does not represent overall resource consumption.

Energy sustainability results calculated through the CED indicator show that gasification with ethanol production has the biggest PE return (avoided impacts) of all analysed recovery technologies, while pyrolysis shows the second-best result. Worst results are achieved by electricity-generating technologies, incineration with electricity production, and gasification with electricity production, due to smaller energy conversion efficiency. The biggest PE return of gasification with ethanol production comes from RES, especially biomass, with over 50% of the overall contribution. In electricity-generating technologies, the majority of renewable energy impacts/benefits are directly dependent on RES share in the electricity mix.

## Conclusion

The plastic waste problem is one of the last identified problems by the EU. Even though the EU is tackling this problem through general WM legislation, and in the last years directly through the legislative framework with a goal of reducing plastic waste generation, problems of plastic are also alleviated through the circular economy and other legislative frameworks which tend to increase the efficiency of resource use and increase the sustainability of overall EU economy. In all of this, the main focus was put on material recovery and the legislative framework for energy recovery is not elaborate enough because of which it classifies all thermochemical conversion technologies in the same category as incineration regardless of sustainability results and what the final products are. This is contrary to other waste recovery legislation which classifies anaerobic digestion of bio-waste as material recovery due to one of the products being a compost-like substance, i.e. not having energy only production. Because of this, this research analysed the environmental, resource, and energy intensity of technologies for energy recovery of plastic waste with a goal of reviewing the existing EU legislation technology classification of thermochemical waste recovery technologies. To give appropriate results, EU legislation on sustainable development was reviewed and the most important impact categories from the legislation aspect were used in this analysis, as well as those

identified by previous research as the most suitable for WM and recovery system analysis and comparison.

From overall results, it can be concluded that pyrolysis of plastic waste and gasification of plastic waste with ethanol production show better results when climate change potential, abiotic depletion potential, and CED impacts are taken into account. Thus, pyrolysis shows a 49/46% decrease in GHG emissions compared to incineration with electricity/CHP production, and gasification with ethanol production GHG emission results is 29/24% lower, respectively. Differences in abiotic depletion results are also substantial in the case of pyrolysis which shows a 143/90% bigger decrease in abiotic depletion, respectively, while in the case of gasification with ethanol production there is an 8% bigger reduction in comparison with incineration with electricity production, while in comparison with CHP production, a 16% smaller reduction is recorded. Large differences can be also seen in the CED category with a 63/55% bigger increase in primary energy return in the case of pyrolysis and 101/91% in the case of gasification with ethanol production, respectively. The only impact indicator that shows better results in the case of incineration-based scenarios when compared to pyrolysis and gasification is AP. Here, results of gasification with ethanol production are 60/32% worse than from incineration with electricity production/CHP production, respectively, while pyrolysis results are the overall worst. Also, regarding direct emissions, all alternative technologies show better results from incineration, and the difference is generated through indirect emissions/savings.

If gasification with electricity production results is looked upon, they are worse than in the case of ethanol generation, and while it shows around 9 to 15% better results than incineration in GHG emissions, results for abiotic depletion are 14 to 33% worse, and in the case of CED 19 to 20% worse than in the case of incineration. On the other hand, cement kiln CED results show less than half of primary energy recovery than gasification with ethanol production and its result is a little better when compared to pyrolysis, its energy recovery is almost on par with other incineration-based scenarios. In the ARD category, it shows second best results, with the only pyrolysis ahead of it and other technologies' results lagging around 40% and more behind its results. On the other hand, the AP category shows that cement kilns can lead to the largest decrease in acidification-related emissions, and in the case of climate change results, it is the only analysed solution that shows a decrease in GHG emissions. But, when taking into account these results, it should be noted that cement kiln results have the widest spread between 5% Percentile and 95% Percentile results.

Presented results show that the environmental impact of a specific technology is largely dependent on the final products which are placed on the market and thus the sustainability of products it replaces. Thus pyrolysis can be considered

largely superior to incineration regarding a large number of EU directives and can help in meeting the goals regarding the establishment of the circular economy, sustainable development, decrease resource use, imports, and climate impacts, as well increase in the security of supply. All of this can also be concluded for gasification with ethanol production, even if ARD results are only, on average, on par with incineration-based technologies. It is because the ARD impact category does not take into account, not depletable resources, such as RES, which are important when conducting sustainability analysis from the legislation point of view. Here, CED impact category proved to be important as it takes into account the consumption of all resources, including RES, and thus complements the results of the ARD impact category. Because of this, it can be concluded that CED is not only the go-to single score impact assessment indicator for benchmarking WM systems, as is concluded in previous research but also an important indicator for sustainability analysis and comparison from the legislation point of view.

The only area where these two technologies are not superior is the air pollution in a form of AP. Even though the reduction of AP-related emissions is larger for incineration-based technologies at this point, these results are strongly linked to the electricity and heat market energy mix and with increased RES share it can be expected that these results will also shift towards pyrolysis and gasification technologies. This is most pronounced in electricity-producing technologies as its market mix quickly is changing towards greater use of RES and is less pronounced in heat generation as district and industrial heating systems transition to other sources of heat (such as electricity or waste heat) much slower. Other recovery technologies are connected to the substitution of final products which production routes are not expected to drastically change in the next decades.

Even though incineration is a less sustainable solution, co-incineration in a cement kiln can be a preferred solution. Here, plastic waste substitutes for coal and petroleum coke which are the most environmentally unsustainable fuels. By doing this, co-incineration of plastic waste becomes the most sustainable and preferred option from the EU legislation standpoint when compared to all other analysed plastic WM solutions.

This analysis provides levelized results for environmental and resource sustainability for MPW recovery technologies in legislatively most important areas. Based on the presented results, it can be concluded pyrolysis and gasification technologies for the treatment of MPW can lead to lower environmental impacts when compared with plastic waste incineration and can help the EU to reach sustainable development goals. This conclusion also answers the research question. These conclusions are viable now, but also in the foreseeable future as the sustainability of electricity and

heat generating technologies is expected to decrease with the meeting of EU RES targets. But before building new treatment facilities dedicated to waste treatment, possibilities for (partial) substitution of less environmentally sustainable fuels in other facilities need to be looked upon, which could lead to even better results from the legislation and sustainability standpoints. By looking upon all these findings which are obtained through legislative recognized approach, it can be also concluded that current views on dedicated, but also not dedicated, thermochemical recovery technologies need to be re-examined and EU institutions need to be encouraged to put the effort in revising EU legislation regarding classifying and ranking of different thermochemical process based recovery technologies taking into consideration type of final products and the final impacts of such production, which also represents a confirmation of the established hypothesis. This conclusion is backed up by the fact that the majority of alternative thermochemical conversion technologies products can be used as inputs in other industries, like pyrolysis oil (which can be used for petroleum substitution) and ethanol, and do not need to be strictly used as fuels (i.e. energy vectors). Thus, the same rezoning for legislation changes can be used as the ones used for classifying anaerobic digestion of bio-waste in the recycling category.

In the future work, this analysis will be expanded with sensitivity analysis which analyse the impact of changes in energy mixes on the results as well as broaden to include economic assessment which also makes one of the important pillars in decision-making.

## Appendix

### Gathered data for modelling of LCI datasets for pyrolysis and gasification

As there were no LCI data representing gasification and pyrolysis technologies in available LCI databases, LCI sets had to be modelled from the beginning. As for legislation making, average data for the specific sector/industry and activity/product should be used and not specific cases which could represent extremes instead of average situation, an extensive literature review of used pyrolysis and gasification technologies for the treatment of plastic waste is conducted and all available technology (technical, input/output and emissions) data on these plants/technologies are gathered and presented in Tables 7, 8, 9 and 10. In these tables, all available data from the cited literature are summarized and encompasses data for 42 individual plants for thermochemical conversion of plastic waste, plastic waste mixtures, and wastes that contain plastic in a significant proportion. The presented data are only adapted from the literature data in

Table 7 Technology data for the formation of LCI dataset—Pyrolysis of plastic waste 1

Reference	Units	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)
Vendor / Technology		Agillyx	Envion	Climax	JB1	H. Smart	VeBa
Location		Tigard, OR	Derwood, MD	Fairfax, SC	Niagara Falls, NY		Bottrop, Germany
Method of Depolymerization/ Feed Process							
Design capacity	tonnes per day	9,071,847.4	26,308,357.46	18,143,694.8	18,143,694.8	48,080,791.22	581,505,418.3
Feedstock requirements		industry standard, grinding/shredding	feedstock is chipped to 1.5 inches or smaller	chipped and shred	shredding or pre-melting		polyeffins
Type of Feedstock (% compositions, if available)		PET, HDPE, PVC, LDPE, PP, PS, other plastics	PET, HDPE, PVC, LDPE, PP, PS, other plastics	100% plastics	HDPE, LDPE, PP		
Contamination limits			PET, PVC in small amounts				
Inorganic matter of feedstock	<%	100	100		5		4.5
Moisture content of feedstock	<%		2	0–5	10		
Energy recovery efficiency	%	82–85	30–80 (62)	75	92		
Heat for drying							
Input							
Tonnage of feedstock	kWh/wet tone dry tonne per day	9,071,847.4	26,308,357.46	18,143,694.8	18,143,694.8		
Power consumption / parasitic load	KWh/dry tone		529,109,429.2	992,080,179.8	0,330,693,393		220,462,262.2
Other inputs (e.g., water, oxygen, etc.)			826,733,483.2				
	Oxygen					trade secret	0.4
	Catalysts and chemicals						
	CaO						0.00005
	Ammonia						
	Sand						
	Hydrogen						1
	E-Gas						11

Table 7 (continued)

Reference	Units	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)
	Nitrogen					
	NaOH					
	HCl					1
	Water		417.2702222	834.5404443	125.1810666	
	Carbon					
	Air					
	Cooling water					
	Natural gas					0.001293219
Supplemental fuel use					0.009699145	
	Off-gass					
	Naphtha					
	Steam					2.26313E-05
	Heat input					
	Heat input startup					
	Syngas					
Output	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)				0.064660969	
	Synthetic crude oil		11.96227928	9.699145366		64.6609691
	Heavy fraction (waxes)		1051.52096	876.2674945		
	Light fraction (liquid)			150–200		0.015
	Gas fraction		100–250			
	Nitrogen					
	Petcoke					
	Gasoline			11.5		10

Table 7 (continued)

Reference	Units	(RTI, 2012)	(Tukker et al., 1999)	(Perugini et al., 2005)	(Perugini et al., 2005)	(Tsamis et al., 2013)	(Tsamis et al., 2013)
Vendor / Technology		BP	BP Chemicals	BP process	Vebe process	Agilyx	Agilyx
Location			Grangemouth				Portland, OR
Method of Depolymerization/ Feed Process							thermal pyrolysis
Design capacity	tonnes per day					43.54486752	27.2155422
Feedstock requirements			size reduction and removal of most non-plastic materials				shredded, granulated, and pelletized
Type of Feedstock (% compositions, if available)			Polyolefins: 80 (min. 70) wt% PS: 15 (max. 30) wt%	Polyolefines	Polyolefines	HDPE, LDPE, PP, other plastic, small amounts of PET	PET, HDPE, PVC, LDPE, PP, PS, other plastics
Contamination limits			PET: 3 (max. 5) % PVC: 2 (max. 4) wt%	max 4% contaminants, 4.5% ash, 2.5% chlorine, and 1% moisture			
Inorganic matter of feedstock	< %			0.5–1	1		
Moisture content of feedstock	< %						
Energy recovery efficiency	%	80		85			
Heat for drying	kWh/wet tone						
Input	dry tonne per day				1	43.54486752	27.2155422
	Tonnage of feedstock						
	Power consumption / parasitic load				58.8888936	266.6666688	29.85426467
	KWh/dry tone	0.033069339					35.82511761
	KW/dry tonne	60					

Table 7 (continued)

Reference	Units	(RTI, 2012)	(Tukker et al., 1999)	(Perugini et al., 2005)	(Perugini et al., 2005)	(Tsiamis et al., 2013)	(Tsiamis et al., 2013)
Other inputs (e.g., water, oxygen, etc.)	Oxygen						
	Catalysts and chemicals						+
	CaO	kg/dry tonne		46	1		
	Ammonia	kg/dry tonne	0.3				
	Sand	kg/dry tonne	0.000002		8.5		
	Hydrogen	kg/dry tonne				11	
	E-Gas	kg/dry tonne					
	Nitrogen	kg/dry tonne					+
	NaOH	kg/dry tonne					
	HCl	kg/dry tonne					
	Water	l/dry tonne	1669.080889		2000		+
	Carbon						
	Air	kg/dry tonne					+
	Cooling water	l/dry tonne		40,000			
Supplemental fuel use	Natural gas	MWh/dry tonne			1.283333344		86.93129628 start up
	Off-gass						recycled
	Naphtha	MWh/dry tonne			0.036388889		
	Steam	MWh/dry tonne				0.031111111	
	Heat input	tonne/tonne		1.2			
	Heat input startup	KWh/dry tonne					
		MWh/dry tonne					



Table 7 (continued)

Reference	Units	(RTI, 2012)	(Tukker et al., 1999)	(Perugini et al., 2005)	(Perugini et al., 2005)	(Tsiamis et al., 2013)	(Tsiamis et al., 2013)
Output	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Syngas					
		MWh/dry tonne					
	Synthetic crude oil	MWh/dry tone				7.712505307	7.104148722
	Heavy fraction (waxes)	kg/dry tonne			822	780.1788764	800
	Light fraction (liquid)	l/dry tonne					911.3181943
	Gas fraction	kg/dry tonne	510	448			
		kg/dry tonne	0.00005	265			
		kg/dry tonne	0.000035	147	90	99.7903214	100
	Nitrogen	MWh/dry tonne				1.265958317	2.644223686
	Petcoke	MWh/dry tonne				0.07407205	
	Gasoline	kg/dry tonne				27.2155422	
Vendor / Technology	Units	Agillyx	Envion	Climax	JB1	H. Smart	Veba
Output	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	kg/dry tonne				750	
		l/dry tonne					
		MWh/dry tonne					
	CaO/CaCl <sub>2</sub>	kg/dry tonne					
	Sand	kg/dry tonne					
	Heat	MWh/dry tonne					
	CaCl <sub>2</sub>	kg/dry tonne					1
	Off-gass	kg/dry tonne					
	HCl	kg/dry tonne					

Table 7 (continued)

Vendor / Technology	Units	Agillyx	Environ	Climax	JB1	H. Smart	Veba
Residuals (e.g., ash, char, slag, etc.)	kg/dry tonne	80			68		
Char							
Solid residues	MWh/dry tonne						
Wax	kg/dry tonne		80				
Spent catalyst and chemicals	l/dry tonne						
Catalys and sludge	kg/dry tonne				trade secret	30	
Spent SCR catalyst	MWh/dry tonne					0.1	
Inorganic sludge	kg/dry tonne						
Residue to incineration	kg/dry tonne		150				1
Non-hazardous solid waste	kg/dry tonne				2.5	0.005	10
Waxy filter to incineration	kg/dry tonne						
Heat losses	MWh/dry tonne						
Water losses	l/dry tonne				125.1810666	1669.080889	
Air Emissions Data							
PM	kg/dry tonne		negligible		0.019		
	daily average						
	half hourly average						
Carbon Dioxide—Fossil (CO <sub>2fossil</sub> )	kg/dry tonne						
CO <sub>2</sub>	kg/dry tonne	481	3.7–9.25	250		450	
Methane (CH <sub>4</sub> )	kg/dry tonne		13–32.5				
HCl	kg/dry tonne				0.00015		
HF	periodic over min 1-h period						
	periodic over min 1-h period						

Table 7 (continued)

Vendor / Technology	Units	Agillyx	Environ	Climax	JB1	H. Smart	Veba
Hydrocarbons	kg/dry tonne			4	0.00017		2
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne	minimum			0.007		
	ppm						
	periodic over min 1-h period						
Nitrous Oxide (N <sub>2</sub> O)	kg/dry tonne	minimum		minimal	0.15		
NOx expressed as NO <sub>2</sub>	kg/dry tonne	0.8	18.1–45.25	minimal	1.205	0.1	
	ppm						
	daily average						
	half hourly average						
Carbon monoxide (CO)	kg/dry tonne	0.5	1.8–4.5	minimal	0.145	0.3	
	ppm						
	daily average						
	half hourly average						
TOC	mg/mm <sup>3</sup>						
	daily average						
	half hourly average						
	daily average						
	half hourly average						
Mercury (Hg)	kg/dry tonne		1.7637E-11				
	periodic over min 1-h period						
Lead (Pb)	kg/dry tonne		0.0001		0.01		
Cadmium (Cd)	mg/mm <sup>3</sup>						
	periodic over min. 30 min period						
VOC	kg/dry tonne	0.8	negligible		0.0085	0.1	
HAP	kg/dry tonne				0.00017		
Dioxins and furans	periodic over min 1-h period						
NH <sub>3</sub>	kg/dry tonne						0.005

Table 7 (continued)

Vendor / Technology	Output	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Diesel	Units	BP	BP Chemicals	BP process	Veba process	JBI Inc.'s "Plastic2Oil" Process	Agilyx
				kg/dry tonne						
				l/dry tonne						
				MWh/dry tonne						
			CaO/CaCl <sub>2</sub>	kg/dry tonne		57				
			Sand	kg/dry tonne		76				
			Heat	MWh/dry tonne			4.1			
			CaCl <sub>2</sub>	kg/dry tonne						
			Off-gass	kg/dry tonne						
			HCl	kg/dry tonne			5			100
			Char	kg/dry tonne						
		Residuals (e.g., ash, char, slag, etc.)		MWh/dry tonne						1.654268128
			Solid residues	kg/dry tonne		200				
			Wax	l/dry tonne						
			Spent catalyst and chemicals	kg/dry tonne						
			Catalys and sludge	kg/dry tonne						
				MWh/dry tonne						
			Spent SCR catalyst	kg/dry tonne						
			Inorganic sludge	kg/dry tonne						
			Residue to incineration	kg/dry tonne				66		
			Non-hazardous solid waste	kg/dry tonne						50
			Waxy filter to incineration	kg/dry/tonne	0.000015					
		Heat losses		MWh/dry tonne						
		Water losses		l/dry tonne						
		Air Emissions Data								

Table 7 (continued)

Vendor / Technology	Units	BP	BP Chemicals	BP process	Veba process	JBI Inc.'s "Plastic2Oil" Process	Agilyx
PM	kg/dry tonne						
	daily average						
	half hourly average						
Carbon Dioxide—Fossil (CO <sub>2(fossil)</sub> )	kg/dry tonne						
CO <sub>2</sub>	kg/dry tonne		345				
Methane (CH <sub>4</sub> )	kg/dry tonne						
HCl	kg/dry tonne						
	periodic over min 1-h period						
HF	mg/mm <sup>3</sup>						
	periodic over min 1-h period						
Hydrocarbons	kg/dry tonne				2.23		
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne	0.0000005		2			0.02
	ppm						
	periodic over min 1-h period						
Nitrous Oxide (N <sub>2</sub> O)	kg/dry/tonne						
NOx expressed as NO <sub>2</sub>	kg/dry tonne	0.0000001		0.3			15.1
	ppm						
	daily average						
	half hourly average						
Carbon monoxide (CO)	kg/dry tonne						3.1
	ppm						

Table 7 (continued)

Vendor / Technology	BP	BP Chemicals	BP process	Veaba process	JBI Inc.'s "Plastic2Oil" Process	Agilyx
	Units					
TOC	daily average					
	half hourly average	mg/mm <sup>3</sup>				
	daily average	mg/mm <sup>3</sup>				
	half hourly average	mg/mm <sup>3</sup>				
Mercury (Hg)		kg/dry tonne				
	periodic over min 1-h period	mg/mm <sup>3</sup>				
Lead (Pb)		kg/dry tonne				
Cadmium (Cd)	periodic over min. 30 min period	mg/mm <sup>3</sup>				
VOC		kg/dry tonne				
HAP		kg/dry tonne				
Dioxins and furans	periodic over min 1-h period	mg/mm <sup>3</sup>				
NH <sub>3</sub>		kg/dry tonne		0.006		

**Table 8** Technology data for the formation of LCI dataset—Pyrolysis of plastic waste 2

Reference	Units	(Haig et al., 2013)	(Haig et al., 2013)	(ORC, 2015)	(ORC, 2015)	(ORC, 2015)
Vendor / Technology		Climax Global Energy Inc Barnwell County, SC	Pyrolysis	Catalytic depolymerisation	Cynar Bristol, UK	Golden Renewables Yonkers, NY, USA
Location						Salt Lake City, UT, USA
Method of Depolymerization/ Feed Process				Thermal Depolymerization Continuous Feed		Catalytic Depolymerization Continuous Feed
Design capacity	tonnes per day	9,071,847.4	76.8	76.8	18,143,694.8	4,535,923.7
Feedstock requirements		shredding	drying	drying		cleaning, drying, shredding
Type of Feedstock (% compositions, if available)		MPW	MWP	MPW	HDPE, LDPE, PP, PS	PVC, LDPE, PP, PS, other plastics
Contamination limits					PVC: 0% PET: 2%	<40% PET+PVC
Inorganic matter of feedstock	< %					
Moisture content of feedstock	< %					
Energy recovery efficiency	%		126	126	96	
Heat for drying	kWh/wet tonne					
Input	dry tonne per day		72.7296	72.7296		19,595,190.38
	Power consumption / parasitic load		16,499,472.02	16,499,472.02	+	211,982,944.4
	Other inputs (e.g., water, oxygen, etc.)					
	Catalysts and chemicals					+ optional
	CaO					
	Ammonia					
	Sand					
	Hydrogen					
	E-Gas					

Table 8 (continued)

Reference	Units	(Tsiamis et al. 2013)	(Haig et al. 2013)	(Haig et al. 2013)	(ORC, 2015)	(ORC, 2015)	(ORC, 2015)
Supplemental fuel use	Nitrogen						
	NaOH	kg/dry tonne					
	HCl	kg/dry tonne					
	Water	l/dry tonne					6018.320512
	Carbon						
	Air	kg/dry tonne					
	Cooling water	l/dry tonne					
	Natural gas	MWh/dry tonne					
	Off-gas						
	Naphtha	MWh/dry tonne					
Output	Steam	l/dry tonne					
		MWh/dry tonne					
		tonne/tonne					
	Heat input	KWh/dry tonne	127.7719113				
	Heat input startup	MWh/dry tonne	1.583949314	0.844772967			
	Syngas	MWh/dry tonne					
	Synthetic crude oil	MWh/dry tonne	9.926082365	7.814149947			
		kg/dry tonne		650.4751848			
		l/dry tonne					1043.175555
		kg/dry tonne					792.8134221
Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Heavy fraction (waxes)						
	Light fraction (liquid)	kg/dry tonne					
	Gas fraction	kg/dry tonne	150				150
		MWh/dry tonne	1.939829073				
	Nitrogen	kg/dry tonne					
	Petcoke	MWh/dry tonne					
		kg/dry tonne					
	Gasoline	kg/dry tonne					



Table 8 (continued)

Reference	Units	(ORC, 2015)	(S.C.Inc, 2018)	(Fiyga et al. 2018)	(Yu et al., 2018)	(Rodriguez et al., 2018)	(ACC, 2017)
Vendor / Technology		Vadxx	Sustante Technologies	Pyrolysis system	R-ONETM (Regenerated Oil & New Energy)	NRP Pyrolysis Process	Comparison of emissions
Location		Akron, OH	Sherwood, Canada		Hukou, Taiwan		
Method of Depolymerization/ Feed Process		Thermal Depolymerization Continuous Feed					
Design capacity	tonnes per day	54.4310844	10	2.4	2	9.0718474	
Feedstock requirements			cleaning, shredding				
Type of Feedstock (% compositions, if available)		HDPE, LDPE, PP, PS, other plastics, Tires, EPDM rubber	HDPE, LDPE, PP, PS	PE, PP, PS	85% (PP+PE+PS) 15% (ABS+PET+PVC, other)	MPW (60% PP, 40% PE)	MPW
Contamination limits							
Inorganic matter of feedstock	< %						
Moisture content of feedstock	< %						
Energy recovery efficiency	%						
Heat for drying	kWh/wet tonne						
Input							
Tonnage of feedstock	dry tonne per day	54.4310844	10				
Power consumption / parasitic load	KWh/dry tonne	967.2781753			250		
Other inputs (e.g., water, oxygen, etc.)	KW/dry tonne				39.35		
Oxygen	kg/dry tonne						
Catalysts and chemicals	kg/dry tonne						
CaO	kg/dry tonne						
Ammonia	kg/dry tonne						
Sand	kg/dry tonne						
Hydrogen	kg/dry tonne						
E-Gas	kg/dry tonne						
Nitrogen	kg/dry tonne						
NaOH	kg/dry tonne						
HCl	kg/dry tonne						
Water	l/dry tonne			50			
Carbon	kg/dry tonne						
Air	kg/dry tonne						
Cooling water	l/dry tonne			2884.7			
						45.359237	
					0.4		

Table 8 (continued)

Reference	Units	(ORC, 2015)	(S.C.Inc, 2018)	(Fyga et al. 2018)	(Yu et al., 2018)	(Rodriguez et al., 2018)	(ACC, 2017)
Supplemental fuel use	Natural gas		start up				
	Off-gass	MWh/dry tonne					
	Naphtha	MWh/dry tonne			+		
	Steam	l/dry tonne			16.07		
	Heat input	MWh/dry tonne					
	Heat input start up	tonne/tonne		411.6			
	Syngas	KWh/dry tonne					
	Synthetic crude oil	MWh/dry tonne					
	hydrogen,	MWh/dry tonne					8.186078689
	electricity,steam)	kg/dry tonne					598
Output	Heavy frac-tion (waxes)	l/dry tonne	876.2674665				738.5682932
	Light fraction (liquid)	kg/dry tonne					
	Gas fraction	kg/dry tonne					
	Nitrogen	MWh/dry tonne	175				312.0715506
	Petcoke	kg/dry tonne					2.149977223
	Gasoline	MWh/dry tonne					45.359237
		kg/dry tonne					
		kg/dry tonne					
		kg/dry tonne					
		kg/dry tonne					
Vendor/Technology	Units	Climax Global Energy Inc	Pyrolysis	Catalytic depoly-merisation	Cynar	Golden Renewables	PK Clean
Output	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	kg/dry tonne		718.0570222			
	Diesel	l/dry tonne					
	CaO/CaCl <sub>2</sub>	MWh/dry tonne	297.9309481				
	Sand	kg/dry tonne					
	Heat	kg/dry tonne					
	CaCl <sub>2</sub>	MWh/dry tonne					
	Off-gass	kg/dry tonne					
	HCl	kg/dry tonne					
	Char	kg/dry tonne	100			50	
	Residuals (e.g., ash, char, slag, etc.)	kg/dry tonne			102.4287223		
	MWh/dry tonne	0.821194308	0.422386484				

Table 8 (continued)

Vendor/Technology	Units	Climax Global Energy Inc	Pyrolysis	Catalytic depolymerisation	Cynar	Golden Renewables	PK Clean
Solid residues	kg/dry tonne						
Wax	l/dry tonne	578,3365464					
Spent catalyst and chemicals	kg/dry tonne			199,5776135			
Catalys and sludge	kg/dry tonne			1,795,142,555			
Spent SCR catalyst	MWh/dry tonne						
Inorganic sludge	kg/dry tonne						
Residue to incineration	kg/dry tonne						
Non-hazardous solid waste	kg/dry tonne						
Waxy filter to incineration	kg/dry tonne						
Heat losses	MWh/dry tonne		0,950369588				
Water losses	l/dry tonne						
Air Emissions Data							
PM							
	kg/dry tonne				15		
	mg/mm <sup>3</sup>				45		
Carbon Dioxide—Fossil (CO <sub>2(fossil)</sub> )	kg/dry tonne						
CO <sub>2</sub>	kg/dry tonne	279,7378616	58,71172122	56,21787686			
Methane (CH <sub>4</sub> )	kg/dry tonne						
HCl	kg/dry tonne						
	mg/mm <sup>3</sup>				15		
HF	mg/mm <sup>3</sup>				2		
Hydrocarbons	kg/dry tonne						
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne						
	ppm						
	mg/mm <sup>3</sup>						75
Nitrous Oxide (N <sub>2</sub> O)	kg/dry tonne						
NOx expressed as NO <sub>2</sub>	kg/dry tonne						

Table 8 (continued)

Vendor/Technology	Units	Climax Global Energy Inc	Pyrolysis	Catalytic depolymerisation	Cynar	Golden Renewables	PK Clean
Carbon monoxide (CO)	daily average				300		
	half hourly average				600		
		ppm					
	daily average				75		
	half hourly average				150		
	daily average				15		
	half hourly average				30		
	periodic over min 1-h period				0.05		
	periodic over min. 30 min period				0.05		
	periodic over min 1-h period				0.1		
Lead (Pb)	kg/dry tonne						
Cadmium (Cd)	mg/mm <sup>3</sup>						
VOC	kg/dry tonne						
HAP	kg/dry tonne						
Dioxins and furans	mg/mm <sup>3</sup>						
NH <sub>3</sub>	kg/dry tonne						
Vendor/Technology	Units	Vadxx	Sustane Technologies	Pyrolysis system	R-ONETM (Regenerated Oil & New Energy)	NRP Pyrolysis Process	Comparison of emissions
Output	kg/dry tonne		900				
Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)							
Diesel	l/dry tonne						
CaO/CaCl <sub>2</sub>	MWh/dry tonne						
Sand	kg/dry tonne						
Heat	kg/dry tonne			1.156			
CaCl <sub>2</sub>	MWh/dry tonne						
Off-gass	kg/dry tonne			3060			
HCl	kg/dry tonne				88.7		
Char	kg/dry tonne	100					
Residuals (e.g., ash, char, slag, etc.)	kg/dry tonne					52.61671492	
	MWh/dry tonne					0.345936185	

Table 8 (continued)

Vendor/Technology	Units	V <sub>adxx</sub>	Sustane Technologies	Pyrolysis system	R-ONETM (Regenerated Oil & New Energy)	NRP Pyrolysis Process	Comparison of emissions
	kg/dry tonne				104.35		
Solid residues	kg/dry tonne						
Wax	l/dry tonne						
Spent catalyst and chemicals	kg/dry tonne						
Catalys and sludge	kg/dry tonne						
Spent SCR catalyst	MWh/dry tonne						
Inorganic sludge	kg/dry tonne						
Residue to incineration	kg/dry tonne						
Non-hazardous solid waste	kg/dry tonne						
Waxy filter to incineration	kg/dry tonne						
Heat losses	MWh/dry tonne			47			
Water losses	l/dry tonne						
Air Emissions Data							
PM	kg/dry tonne		0.089142857		0.002667139		0.2
	mg/mm <sup>3</sup>						
daily average	mg/mm <sup>3</sup>						
half hourly average	mg/mm <sup>3</sup>						
Carbon Dioxide—Fossil (CO <sub>2fossil</sub> )	kg/dry tonne	0.5867399					
CO <sub>2</sub>	kg/dry tonne						
Methane (CH <sub>4</sub> )	kg/dry tonne						
HCl	kg/dry tonne						
	mg/mm <sup>3</sup>						
periodic over min 1-h period	mg/mm <sup>3</sup>						
periodic over min 1-h period	mg/mm <sup>3</sup>						
Hydrocarbons	kg/dry tonne						
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne		0.011428571		0.0402		0.166666667
	ppm						
periodic over min 1-h period	mg/mm <sup>3</sup>						
Nitrous Oxide (N <sub>2</sub> O)	kg/dry tonne						
NOx expressed as NO <sub>2</sub>	kg/dry tonne		1.659428571		0.01824		0.766666667

Table 8 (continued)

Vendor/Technology	Units	Vadxx	Sustaine Technologies	Pyrolysis system	R-ONETM (Regenerated Oil & New Energy)	NRP Pyrolysis Process	Comparison of emissions
Carbon monoxide (CO)	ppm						
	daily average						
	mg/mm <sup>3</sup>						
	half hourly average						
	mg/mm <sup>3</sup>		0.930285714				0.5333333333
	kg/dry tonne						
TOC	ppm						
	daily average						
	mg/mm <sup>3</sup>						
	half hourly average						
	mg/mm <sup>3</sup>						
	daily average						
	mg/mm <sup>3</sup>						
	half hourly average						
	mg/mm <sup>3</sup>						
Mercury (Hg)	kg/dry tonne						
	periodic over min 1-h period						
	mg/mm <sup>3</sup>						
Lead (Pb)	kg/dry tonne						
	periodic over min. 30 min period						
	mg/mm <sup>3</sup>						
Cadmium (Cd)	kg/dry tonne						
	periodic over min. 30 min period						
	mg/mm <sup>3</sup>		0.121142857				0.3333333333
VOC	kg/dry tonne						
HAP	kg/dry tonne						
Dioxins and furans	mg/mm <sup>3</sup>						
NH <sub>3</sub>	kg/dry tonne						

Table 9 Technology data for the formation of LCI dataset—Gasification of plastic waste 1

Reference	Units	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(Caroline et al., 2010)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Tukker et al., 1999)
Vendor/Technology		Enkhem (Pontotoc)	Ze-gen	Plasco	Alter NRG—integrated gasification combined cycle	Gasification	Gasification and F-T synthesis	Gasification and methanol-to-gasoline synthesis	Gasification and bioconversion to ethanol	Texaco process
Location		Pontotoc, MS	Narragansett Bay, MS	Ottawa, Ontario, Canada						Montebello, California
Design capacity	tonnes per day	299,3709642	68–136	84,36818082	710	76.8	76.8	76.8	76.8	10
Feedstock requirements		Sorting, drying, shredding	Sorting, drying, shredding	elimination of metals, shredding	10 inches size					Shredded or chipped
Type of Feedstock (% compositions, if available)		post-MRF-sorted MSW, industrial waste, construction and demolition waste, treated wood, bagasse, corn stover, wheat straw, rice hulls, wood chips, sawdust, bark, thinning, limbs, needles	95% wood based material, consisting of railroad cross-ties (90%), clean wood waste (5%), nonrecycled source-separated plastics (5%)	Paper and cardboard (24.3%), Plastics (16.2%), Metals (7.2%), Glass (6.1%), Rubber & Leather (3.3%), Textiles (5.9%), Wood (7.4%), Food Scraps (18%), Yard Trimmings (7.3%), Miscellaneous Inorganic Waste (2.2%), Other (2%)	MSW	dried waste plastics	dried waste plastics	dried waste plastics	dried waste plastics	MPW(<10% PVC)
Inorganic matter content of feedstock	< %	15	5							10
Moisture content of feedstock	< %		20			5	5	5	5	5
Efficiency of the electricity generating unit (ICE)	%		85							
Energy recovery efficiency	%	> 72	48	98						
Heat for drying	kWh/wet tonne					126	126	126	126	

Table 9 (continued)

Reference	Units	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(Caroline et al., 2010)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Tukker et al., 1999)
Input										
Tonnage of feedstock	dry tonne per day	299.3709642	68–136	84.36818082	710	72.7296	72.7296	72.7296	72.7296	72.7296
Power consumption / parasitic load	KWh/dry tonne	540.1325424	220.4622622		200	383.3157339	665.2587117			
Other inputs (e.g., water, oxygen, etc.)	kg/dry tonne	723			172	1102.428722	1102.428722	1102.428722	1102.428722	+
Air	kg/dry tonne									
Catalysts and chemicals	kg/dry tonne									
Diesel for preprocessing	l/dry tonne			0.208635111						
Caustic for gas cleaning and cooling	kg/dry tonne			5						
Chemicals, Catalysts, Guard Bed Materials	kg/dry tonne	45.45454545								
Activated Carbon for gas cleaning and cooling	l/dry tonne			0.834540444						
Feldspar for gas cleaning and cooling	kg/dry tonne									
Heat input	kWh/dry tonne				115.2	22.17529039	22.17529039	22.17529039	22.17529039	
Steam	kWh/dry tonne									105.5966209
Coke	kg/dry tonne				38.9					473.0728617
Lignite	kg/dry tonne									
Water	l/dry tonne	6768.123003	2253.2592							



Table 9 (continued)

Reference	Units	(RTI, 2012)	(RTI, 2012)	(Caroline et al., 2010)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Tukker et al., 1999)
	kg/dry tonne						5630.411827	
	kg/dry tonne							
	kg/dry tonne							
Supplemental fuel use	kWh/dry tonne	7.86	1000	43.5				
	kg/dry tonne							
	MWh/dry tonne		439.9458333					
	m <sup>2</sup> /dry tonne							
	kg/dry tonne							
Fuel oil	kg/dry tonne							
	kg/dry tonne							
Electricity	KWh/dry tonne			925–1302/0.907	929			
Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	KWh/dry tonne							
	MWh/dry tonne		29,0974361		8.342133052			
Syngas	kg/dry tonne				1940.865892			
	kg/dry tonne							2333.333333
	Nm <sup>3</sup> /dry tonne							
Steam	MWh/dry tonne		3.233048455		0.739176346	2.111932418	2.111932418	
	kg/dry tonne				954.593453	2864.836325	2813.093981	
Hydrogen	kg/dry tonne							
	kg/dry tonne		280–307.5					
Ethanol	kg/dry tonne						616.6842661	
	MWh/dry tonne						5.068637804	
Methanol	kg/dry tonne							
	kg/dry tonne							
Purge gas	MWh/dry tonne				1.372756072	2.217529039		
	kg/dry tonne				1212.249208	1284.05491		
F-T Liquids	MWh/dry tonne				3.167898627			
	kg/dry tonne				240.7602957			
F-T Waxes	MWh/dry tonne				2.006335797			

Table 9 (continued)

Reference	Units	(RTI, 2012)	(RTI, 2012)	(RTI, 2012)	(Caroline et al., 2010)	(Haig et al., 2013)	(Haig et al., 2013)	(Haig et al., 2013)	(Tukker et al., 1999)	
	kg/dry tonne						154.1710665			
	MWh/dry tonne							4.751847941		
	kg/dry tonne							354.8046463		
Vendor / Technology	Units	Enerkem (Pon-totec)	Ze-gen	Plasco	Alter NRG—integrated gasification combined cycle	Gasification	Gasification and F-T synthesis	Gasification and methanol-to-gasoline synthesis	Gasification and bioconversion to ethanol	Texaco process
Output	Material Byproducts	Reactor off-gas	MWh/dry tonne						2.006335797	
	Residual gas		214						1469.904963	
	Sulphur			1.5						
	Salt			4.5–6.5						
	Slag			12–212						
	Filter cake									
	NaCl									
	HCl									
	Solids									
	Residuals (e.g., ash, char, slag, etc.)	Char	148.66							
	Slag									
	Tar					141.499472	141.499472	141.499472	141.499472	
	Gasifier solid residues		60							
	Spent catalysts and chemicals		1.695							
	Ash					20.06335797	20.06335797	20.06335797	20.06335797	
	Air Pollution Control System residues									
	Inorganic sludge		22.5							
	Gypsum									
	Non-hazardous solid waste		6.5							

Table 9 (continued)

Vendor / Technology	Units	Enerkem (Pon-totoc)	Ze-gen	Plasco	Alter NRG—integrated gasification combined cycle	Gasification and F-T synthesis	Gasification and methanol-to-gasoline synthesis	Gasification and bioconversion to ethanol	Texaco process
Water	kg/dry tonne					675.8183738	644.1393875	5485.744456	
Potable water	kg/dry tonne								
	l/dry tonne			2086.5491–312.519.2944					
Heat losses	MWh/dry tonne				1.267159451	1.478352693	1.478352693	1.900739176	1.267159451
Water losses	l/dry tonne	4172.702222	2086.351111						
	kg/dry tonne								
Air Emissions Data									
PM	kg/dry tonne	0.1765	0.005	0.021–0.022					
PM10	kg/dry tonne			0.00035					
PM2.5	kg/dry tonne								
Carbon Dioxide—Biogenic (CO <sub>2bio</sub> )	kg/dry tonne			233.52					
Carbon Dioxide—Fossil (CO <sub>2fossil</sub> )	kg/dry tonne	201.94	172.5	523.78					
CO <sub>2eq</sub>	kg/dry tonne			220–354		301.1615628	1050.686378	1285.216473	
Methane (CH <sub>4</sub> )	kg/dry tonne	0.945		0.0001					
HCl	kg/dry tonne			0.012–0.01,298					
Hydrocarbons	kg/dry tonne		0.004						
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne	0.093	0.19	0.058–0.086					
Sulphur oxide (SO)	kg/dry tonne			0.000025					
Nitrous Oxide (N <sub>2</sub> O)	kg/dry tonne	0.1975		0.0005					
NOx expressed as NO <sub>2</sub>	kg/dry tonne	0.555	0.095	0.084–0.086					
Carbon monoxide (CO)	kg/dry tonne	0.73	0.065	0.205–0.22					
Mercury (Hg)	kg/dry tonne		0.0000017	0.0000003					
Cadmium (Cd)	kg/dry tonne		0.000000255	0.0000004					
Lead (Pb)	kg/dry tonne		0.0000003595	0.0000005					
VOC	kg/dry tonne	0.45	0.02						
HAP	kg/dry tonne	0.05							

Table 9 (continued)

Vendor / Technology	Units	Enerkem (Pon-totoc)	Ze-gen	Plasco	Alter NRG—integrated gasification combined cycle	Gasification and F-T synthesis	Gasification and methanol-to-gasoline synthesis	Gasification and bioconversion to ethanol	Texaco process
NH <sub>3</sub>	kg/dry tonne								
Dioxins and furans	kg/dry tonne								
Acetaldehyde	kg/dry tonne	0.03							
TNMOc	kg/dry tonne			0.1					
Antimony (Sb)	kg/dry tonne								
Arsenic (As)	kg/dry tonne								
Titanium (Ti)	kg/dry tonne								
Chromium (Cr)	kg/dry tonne								
Iron (Fe)	kg/dry tonne								
Copper (Cu)	kg/dry tonne								
Zinc (Zn)	kg/dry tonne								
Water Emissions Data	kg/dry tonne								
Water Effluent	l/dry tonne	2504—5842							
	kg/dry tonne			1453.05—3594.85					

Table 10 Technology data for the formation of LCI dataset—Gasification of plastic waste 2

Reference	Units	(Tukker et al., 1999)	(Tukker et al., 1999)	(PowerHouse, 2019)	(PowerHouse, 2019)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	
Vendor		SVZ process	Akzo Nobel Stream Gasification Process	POWER HOUSE ENERGY GROUP, DMG	POWER HOUSE ENERGY GROUP, DMG	I	II	III	IV	V	VI
Location											
Design capacity	tonnes per day			25	25						
Feedstock requirements		particle size: 20–80 mm, chlorine content: 2%		Sorting, Drying, Shredding.	Sorting, Drying, Shredding.						
Type of Feedstock compositions, if available)		MPW agglomerate, waste oil	PP, PE, PVC	SRF, plastics, WEE plastics, tyre	SRF, plastics, WEE plastics, tyre	PE—Recycled polyethylene, derived from separate collection of MSW	GS3—Mix of recycled polyolefinic plastics obtained from plastic packaging for food and beverages by means of sorting and washing treatments	Neolite—Mix of plastics obtained from separate collection of plastic post-consumer packaging, but containing also ferrous and non-ferrous metals	Mix of plastics obtained from separate collection	PDF—Mix of different kinds of food packaging, generally consisting of multilayer plastic, paper and aluminium	PDF—Mix of different kinds of food packaging, generally consisting of multilayer plastic, paper and aluminium
Inorganic matter content of feedstock	<%										
Moisture content of feedstock	<%										
Efficiency of the electricity generating unit (ICE)	%										
Energy recovery efficiency	%										
Heat for drying	kWh/wet tonne										
Input	tonnage of feedstock			25	25	1	1	1	1	1	1
	Power consumption / parasite load				115,200						
	Other inputs (e.g., water, oxygen, etc.)		Oxygen		1442.590775						
	Air				2300						
	Catalysts and chemicals										
	Diesel for preprocessing										

Table 10 (continued)

Reference	Units	(Tukker et al., 1999)	(Tukker et al., 1999)	(PowerHouse, 2019)	(PowerHouse, 2019)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)
Caustic for gas cleaning and cooling	kg/dry tonne									
	kg/dry tonne									
Chemicals, Catalysts, Guard Bed Materials	kg/dry tonne									
	l/dry tonne									
Activated Carbon for gas cleaning and cooling	kg/dry tonne					0.508196721	0.491525424	0.5	0.5	0.5
	l/dry tonne									
Feldspar for gas cleaning and cooling	kWh/dry tonne									
	kWh/dry tonne									
Heat input	kWh/dry tonne									
	kWh/dry tonne									
Steam	kg/dry tonne		300							
	kg/dry tonne									
Coke	kg/dry tonne									
	kg/dry tonne									
Lignite	kg/dry tonne		1226.692836							
	l/dry tonne									
Water	kg/dry tonne		7752.698724							
	kg/dry tonne									
Hydrated lime	kg/dry tonne									
	kg/dry tonne					6.557377049	6.440677966	6.477272727	6.40625	6.465517241
Supplemental fuel use	Syngas									
	Natural gas									
Fuel oil	MWh/dry tonne									
	m <sup>2</sup> /dry tonne		98.13542689							
Electricity	kg/dry tonne		39.25417076							
	kWh/dry tonne		621.5243753							
Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	MWh/dry tonne									
	kg/dry tonne									
Steam	Nm <sup>3</sup> /dry tonne		200.1962709							
	MWh/dry tonne									
Hydrogen	kg/dry tonne									
	kg/dry tonne									
Ethanol	MWh/dry tonne									
	kg/dry tonne									
Methanol	kg/dry tonne									
	MWh/dry tonne									
Purge gas	kg/dry tonne									
	kg/dry tonne									

Table 10 (continued)

Reference	Units	(Tukker et al., 1999)	(Tukker et al., 1999)	(PowerHouse, 2019)	(PowerHouse, 2019)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	(Ardolino et al., 2018)	
F-T Liquids	MWh/dry tonne									
	kg/dry tonne									
F-T Waxes	MWh/dry tonne									
	kg/dry tonne									
Gasoline	MWh/dry tonne									
	kg/dry tonne									
Vendor	Units	SVZ process	Akzo Nobel Stream Gasification Process	POWER HOUSE ENERGY GROUP, DMG	POWER HOUSE ENERGY GROUP, DMG	II	III	IV	V	VI
Output	Material Byproducts	Reactor off-gas		4.5968						
		Residual gas								
		Sulphur								
		Salt								
		Slag								
		Filter cake			0.0104					
		NaCl			1.9968					
		HCl	210							
		Solids			4.68					
	Residuals (e.g., ash, char, slag, etc.)	Char								
		Slag	0.883218842							
		Tar								
		Gasifier solid residues								
		Spent catalysts and chemicals								
		Ash	220		146.8852459	100.8474576	82.27272727	35.78125	68.36206897	68.36206897
		Air Pollution Control System residues			7.049180328	6.949152542	6.931818182	6.875	6.982758621	6.982758621

Table 10 (continued)

Vendor	Units	SVZ process	Akzo Nobel Stream Gasification Process	POWER HOUSE ENERGY GROUP, DMG	I	II	III	IV	V	VI
Inorganic sludge	kg/dry tonne									
Gypsum	kg/dry tonne	98.13542689								
Non-hazardous solid waste	kg/dry tonne									
Water	kg/dry tonne									
Potable water	kg/dry tonne									
Heat losses	l/dry tonne									
Water losses	MWh/dry tonne									
Air Emissions Data	l/dry tonne	9715.407262								
PM	kg/dry tonne									
PM10	kg/dry tonne									
PM2.5	kg/dry tonne				0.01	0.012881356	0.066931818	0.01890625	0.061551724	
Carbon Dioxide—Biogenic (CO <sub>2bio</sub> )	kg/dry tonne									
Carbon Dioxide—Fossil (CO <sub>2fossil</sub> )	kg/dry tonne	12,177.26397			2622.95082	2762.711864	2409.090909	2828.125	1922.413793	1922.413793
CO <sub>2eq</sub>	kg/dry tonne									
Methane (CH <sub>4</sub> )	kg/dry tonne									
HCl	kg/dry tonne				0.000180328		0.104204545	0.0053125	0.027327586	0.027327586
Hydrocarbons	kg/dry tonne									
Sulphur dioxide (SO <sub>2</sub> )	kg/dry tonne									
Sulphur oxide (SO)	kg/dry tonne						0.098863636	0.0984375	0.103448276	0.103448276
Nitrous Oxide (N <sub>2</sub> O)	kg/dry tonne									



Table 10 (continued)

Vendor	Units	SVZ process	Akzo Nobel Stream Gasification Process	POWER HOUSE ENERGY GROUP, DMG	I	II	III	IV	V	VI
NOx expressed as NO <sub>2</sub>	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	0.070327869	0.071355932	0.0625	0.06953125	0.066034483	0.066034483
Carbon monoxide (CO)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
Mercury (Hg)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	6.55738E-07	6.77966E-07	1.13636E-06	1.5625E-06	8.62069E-07	8.62069E-07
Cadmium (Cd)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	1.63934E-06	1.69492E-06	2.27273E-05	4.6875E-06	1.72414E-06	1.72414E-06
Lead (Pb)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	0.000245902	0.002305085	0.008522727	0.000328125	0.000724138	0.000724138
VOC	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
HAP	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
NH <sub>3</sub>	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	3.27869E-05	3.38983E-05	3.40909E-05	0.00003125	3.44828E-05	3.44828E-05
Dioxins and furans	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	6.55738E-12	6.77966E-12	6.81818E-12	6.25E-12	3.44828E-12	3.44828E-12
Acetaldehyde	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
TNMOc	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
Antimony (Sb)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	3.27869E-06	1.52542E-05	0.000352273	5.3125E-07	1.12069E-05	1.12069E-05
Arsenic (As)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	6.55738E-07	6.77966E-07	1.13636E-06	1.5625E-06	8.62069E-07	8.62069E-07
Titanium (Ti)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG	3.27869E-06	1.69492E-06	2.27273E-06	0.000003125	2.58621E-06	2.58621E-06
Chromium (Cr)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG			0.000988636	0.00009375		
Iron (Fe)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG			0.004659091	0.0028125	0.001293103	0.001293103
Copper (Cu)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG			0.013068182	0.000171875	2.58621E-05	2.58621E-05
Zinc (Zn)	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG				0.0000625		
Water Emissions Data	kg/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
Water Effluent	l/dry tonne			POWER HOUSE ENERGY GROUP, DMG						
	kg/dry tonne	9630.350915								

a way that they are converted to the metric system to be comparable.

As can be seen, available data from different data sources vary greatly, both in the amount of data and in the form of their presentation. Thus, for the formation of a representable dataset, many data sources are consulted and collected data adapted and averaged to represent the general dataset for analysed technologies. This way, the lack of data from individual data sources can be compensated, as well as errors and inconsistencies in the gathered data.

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**Data availability** All used data and materials are referenced in the manuscript.

**Code availability** Not applicable.

## Declarations

**Conflict of interest** The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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